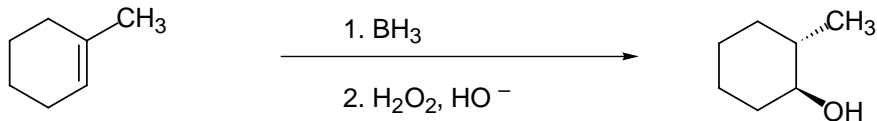


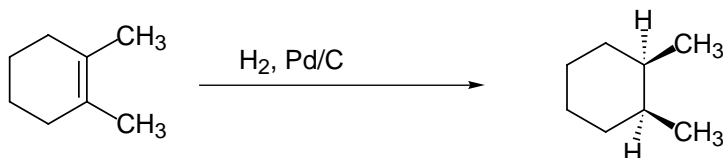
I. REACTIONS: Predict the major organic products of the following reactions.

**INDICATE STEREOCHEMISTRY AS NEEDED.** (4 points each)

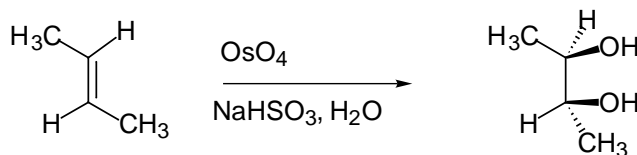
1.



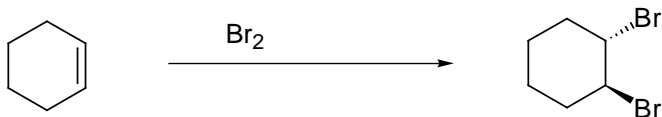
2.



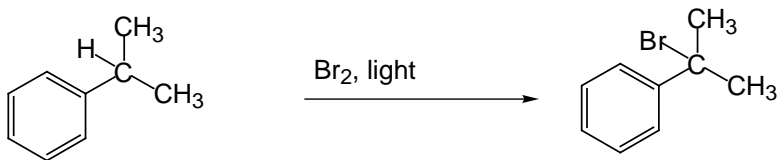
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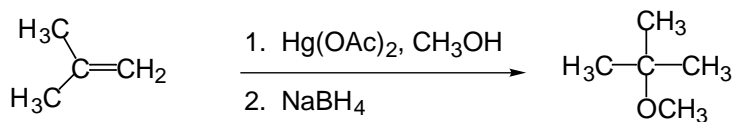
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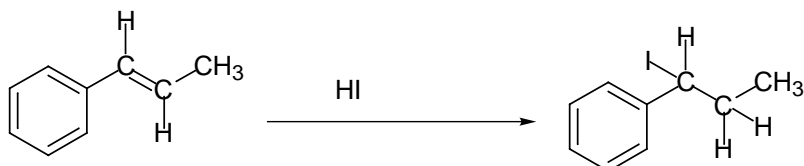
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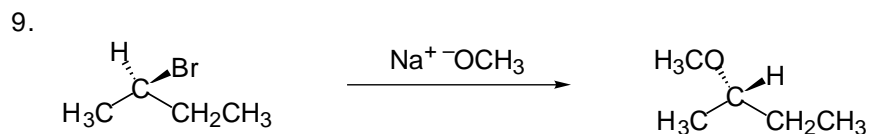
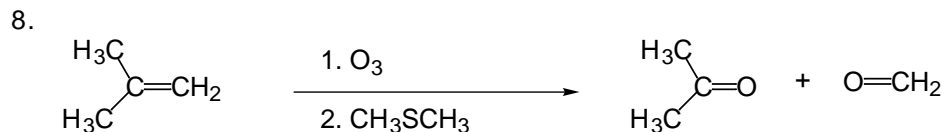


6.

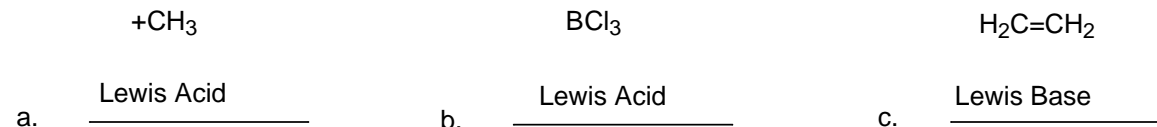


7.



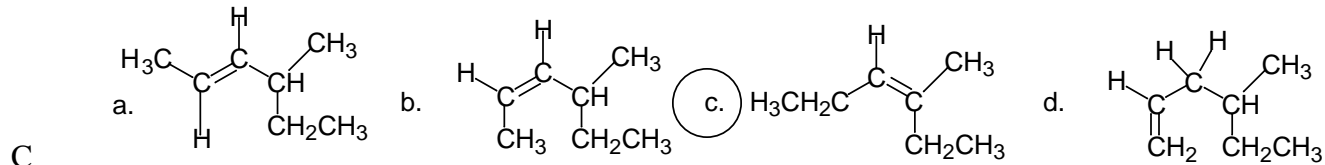


10. Label each of the following as a Lewis acid or Lewis base. 2 points each.

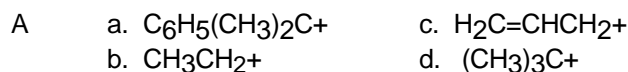


II. Multiple Choice: Circle the best answer (**only one**). (3 points each)

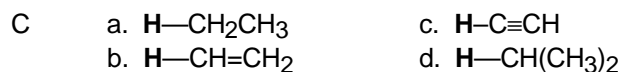
11. Which of the following alkenes has the lowest heat of hydrogenation?



11. Which of the following is the **most** stable carbocation?



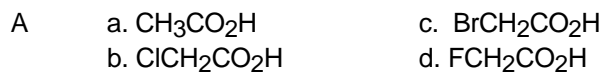
12. Which of the following is the **most** acidic?



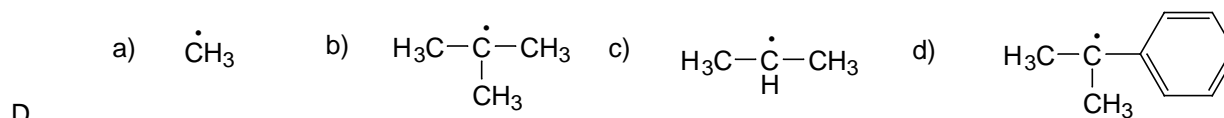
13. Which of the following is the **most** acidic?



14. Which of the following is the **least** acidic?



15. Which of the following free radicals is the **most** stable?



16. List four things which affect acidity. (1 point each)

size of the atom bearing the charge

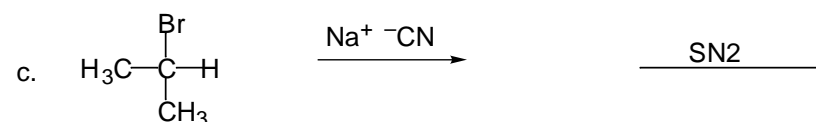
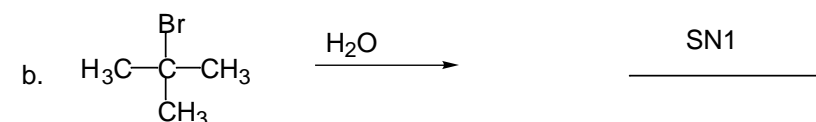
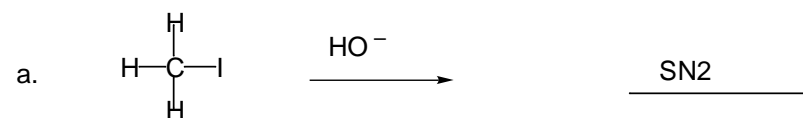
electronegativity of the atom bearing the charge

Resonance

Induction by nearby electron withdrawing groups

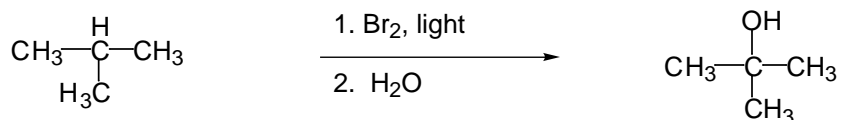
hybndization of the atom bearing the charge

17. Would you expect the following to proceed by SN1 or SN2 mechanism? (2 points each)



Syntheses: Give reagents to carry out the transformations below. (4 points each)

18.



19. Choose from: SN1, SN2, free radical substitution, electrophilic addition, hydrogenation, dihydroxylation, ozonolysis, hydroboration. (2 points each)

stepwise reaction free radical substitution or SN1 or electrophilic addition

stereospecific reaction SN2, or hydroboration or hydrogenation or dihydroxylation

concerted reaction SN2, or hydroboration or hydrogenation or dihydroxylation

chain reaction free radical substitution

first order kinetics SN1

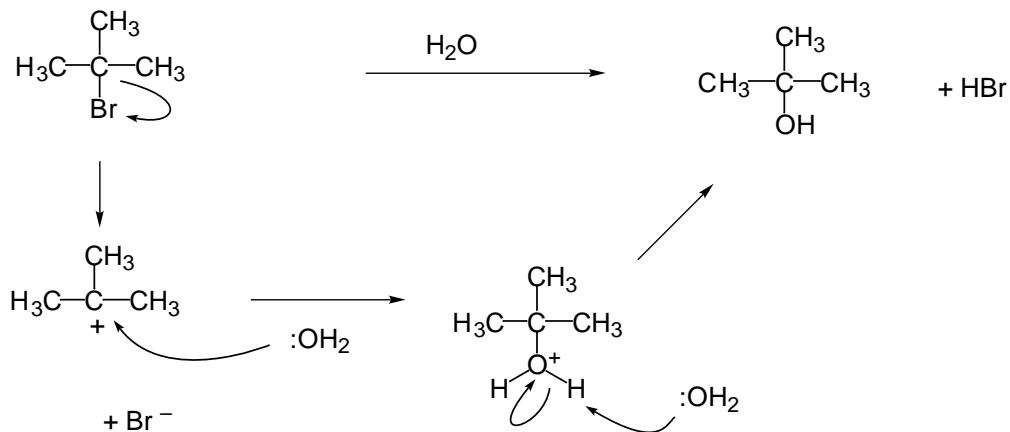
second order kinetics SN2 electrophilic addition

inversion of stereochemistry SN2

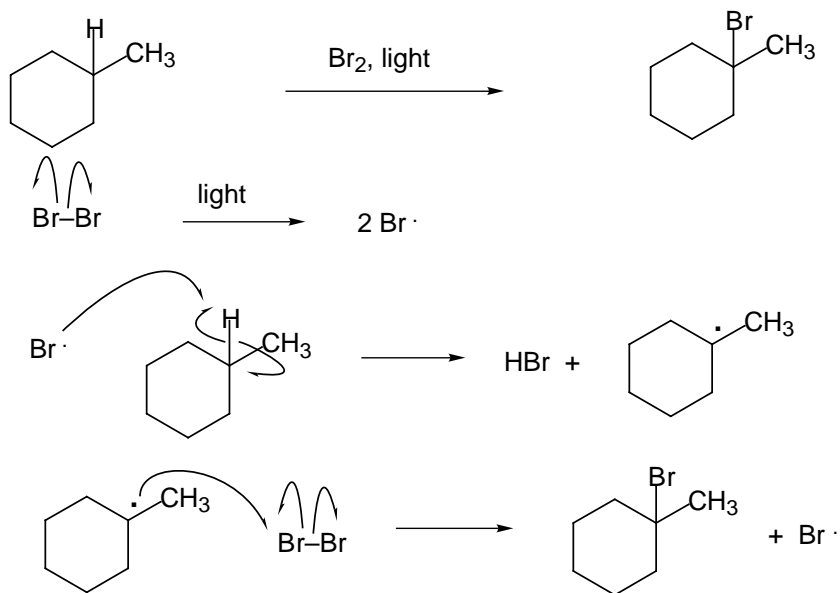
racemization of stereochemistry SN1

V. Mechanisms. Give a stepwise, detailed mechanism with arrows and intermediates for the following reactions. (4 points each)

20.



21.



22.

