

Recent Advances in Rhodium Catalyzed Asymmetric Cascade Reactions

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What Exactly is a Cascade Reaction?

Cascade, Tandem, Domino, Sequential,
Multidirectional, One-pot transformation...

A cascade reaction is any bond forming
reaction or chemical transformation that
generates a reactive intermediate that
undergoes a subsequent intra- or
intermolecular reaction

or more simply...

“Any reaction that generates molecular
complexity in a concise fashion”

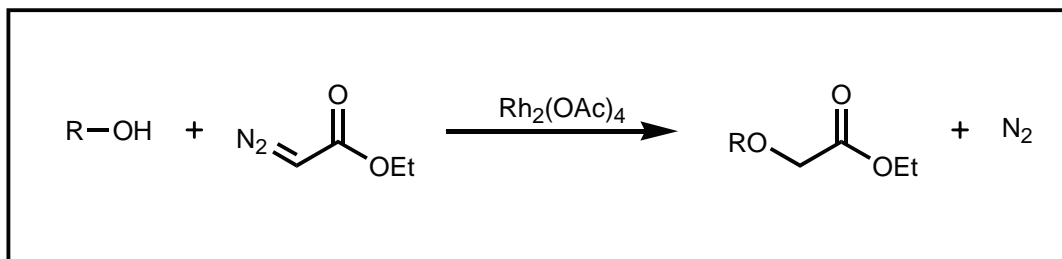
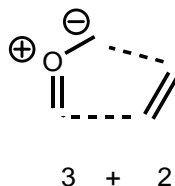
Why is Rhodium Special?

- Many Rhodium catalysts are air/moisture stable
- Metallocarbene formation
 - 1,3 dipole formation
 - C-H activation/insertion
 - Cyclopropanation
- Oxidative coupling
- Cyclopropane/cyclobutane opening
- Decarbonylation

Topics for Today's Discussion

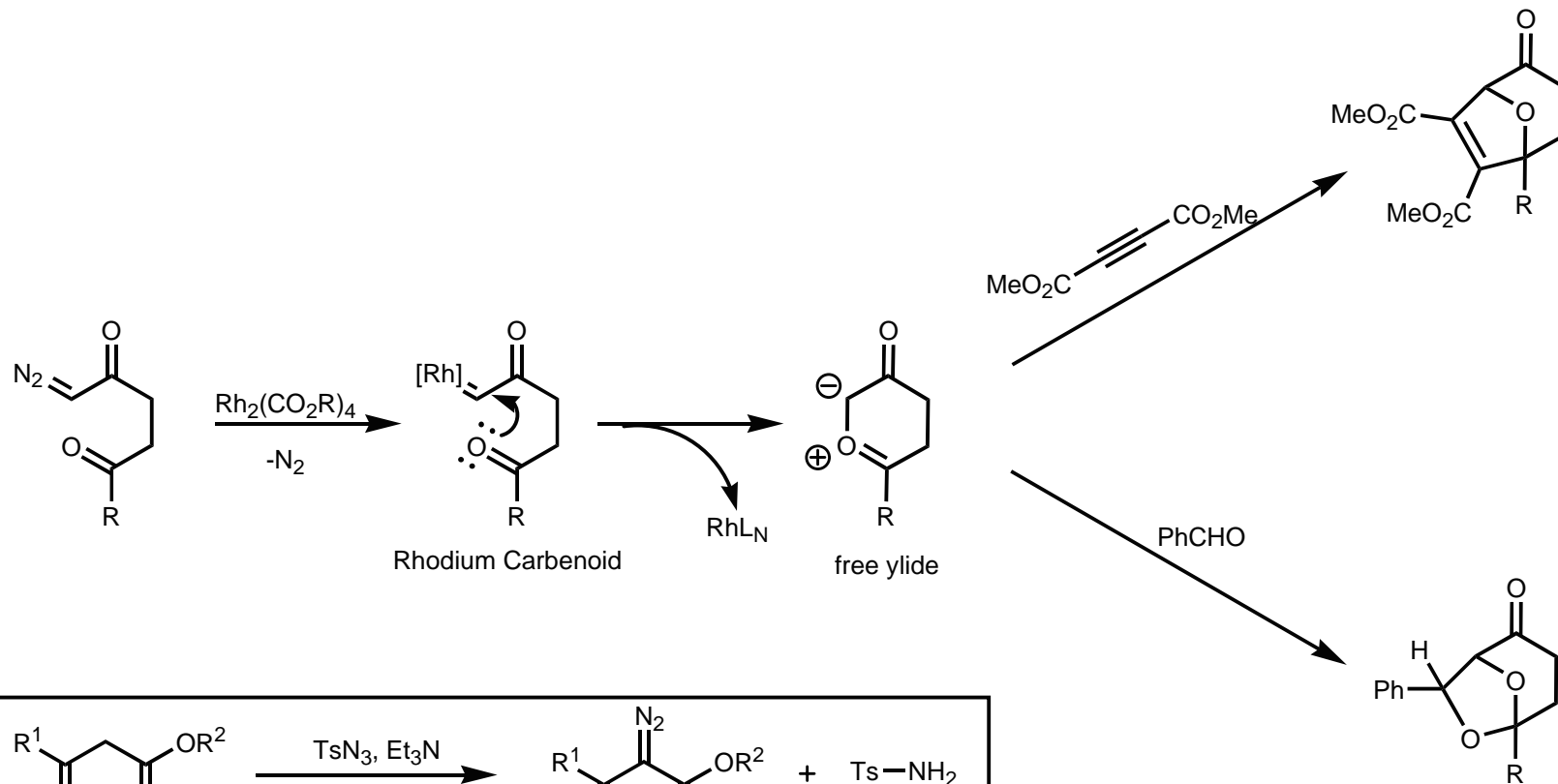
- 1,3-Dipole Generation/[3+2] Cycloadditions
- C-H Activation/Cope Rearrangement
- Formal [5+2] Cycloadditions
- Formal [2+2+1] Pauson-Khand Reactions
- Formal [2+2+2] Cycloadditions

Rhodium Catalyzed [3+2] Dipolar Cycloadditions



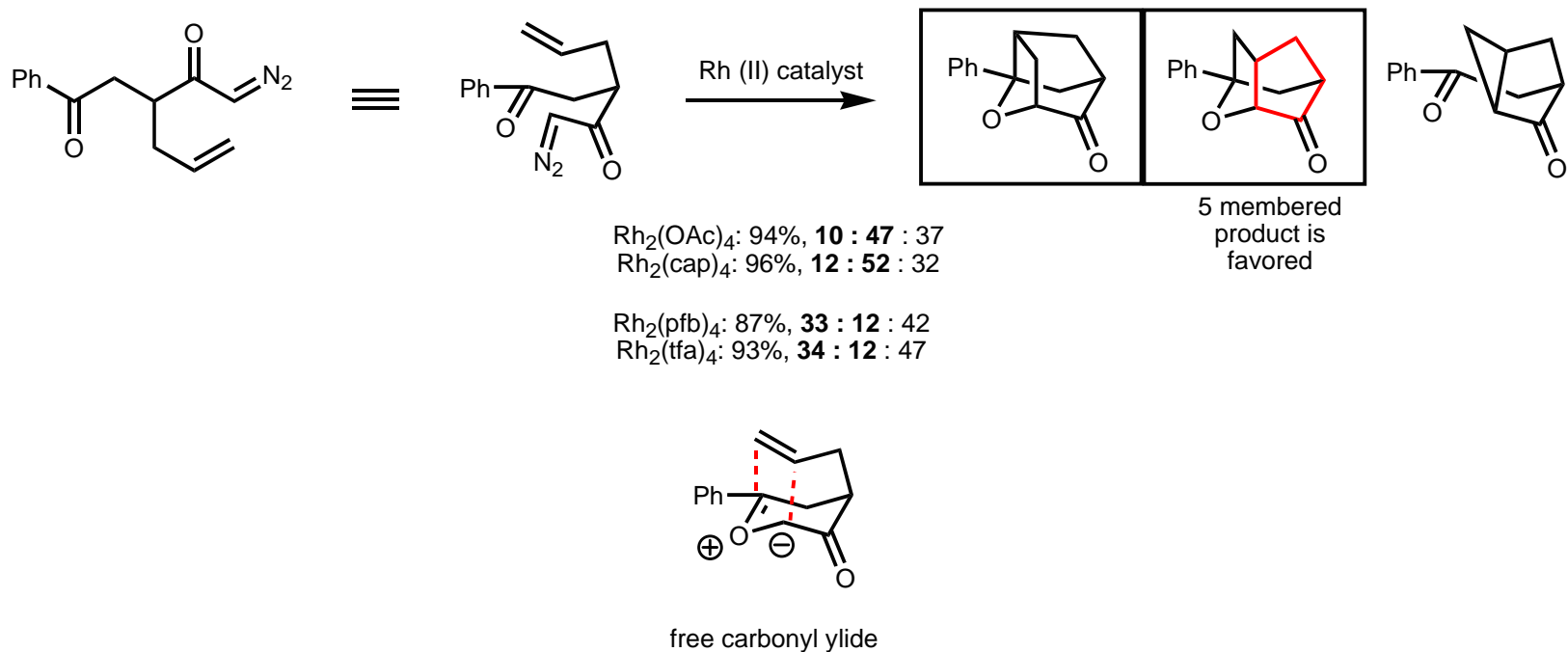
Padwa, A.; Weingarten, M. D. *Chem. Rev.* **1996**, 96, 223. Paulisson, R.; Reimlinger, H.; Hayez, A. J.; Teyssié, P. H. *Tetrahedron Lett.* **1973**, 2233.

Rhodium Catalyzed [3+2] Dipolar Cycloadditions



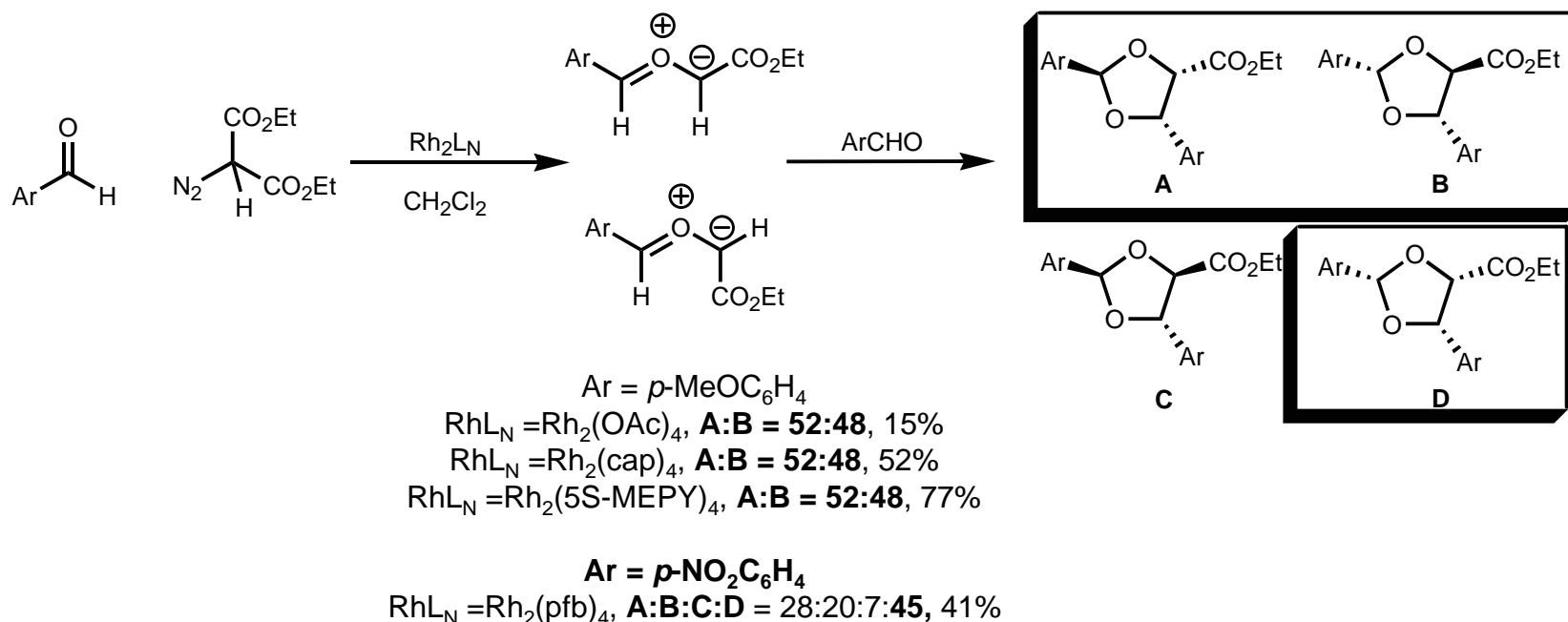
Padwa, A.; Weingarten, M. D. *Chem. Rev.* **1996**, 96, 223. Regitz, M.; Maas, G. *Diazo Compounds: Properties and Synthesis*; Academic Press: Orlando, 1986.

Rhodium Catalyzed Regioselective [3 + 2] Cycloadditions



Changing the catalyst to rhodium(II) trifluoroacetate or to rhodium(II) perfluorobutyrate generated a significant change in the product distribution

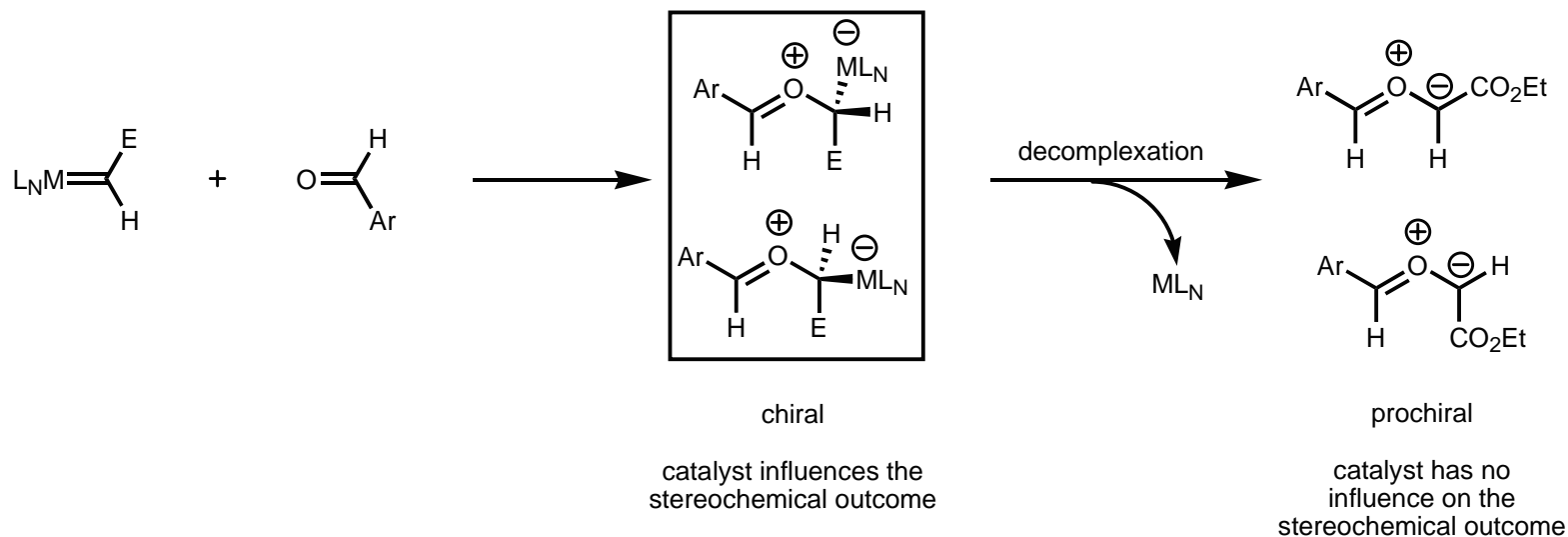
Rhodium Catalyzed Regioselective [3 + 2] Cycloadditions



If a free ylide is formed the catalyst shouldn't influence the regiochemical outcome. Therefore, in the case of fluorinated ligands, perhaps the metal is still bound to the original site of attachment.

Doyle, M. P.; Forbes, D. C.; Protopopova, M. N.; Stanley, S. A.; Vasbinder, M. M.; Xavier, K. R. *J. Org. Chem.* **1997**, *62*, 7210.

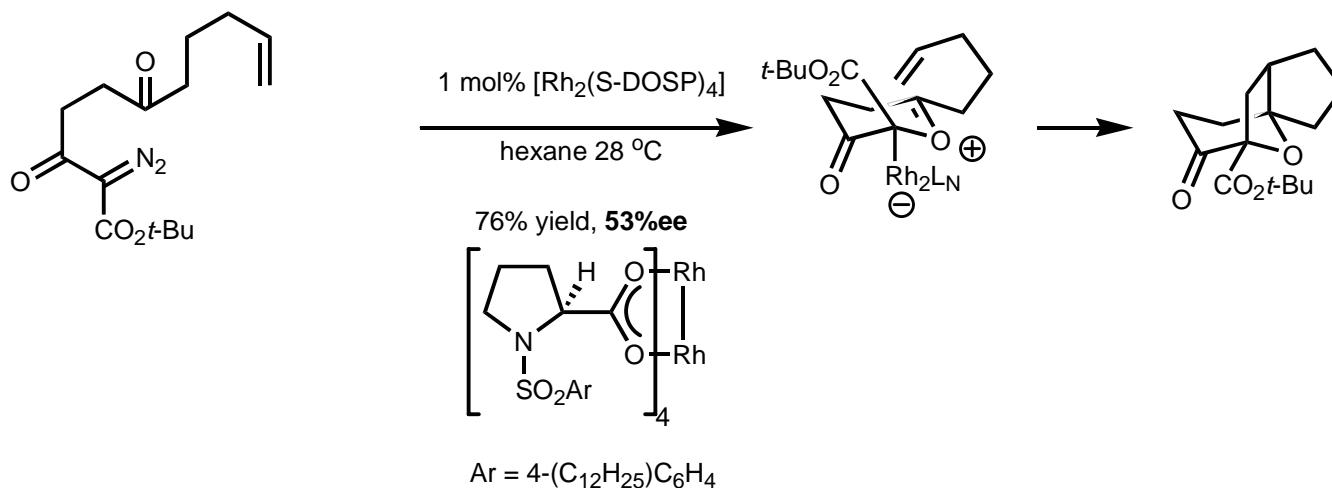
Rhodium Catalyzed Regioselective [3 + 2] Cycloadditions



If the dipole is not fully formed and the metal is still bound to the original site of attachment, the ligands on the metal may be able to provide stereochemical communication

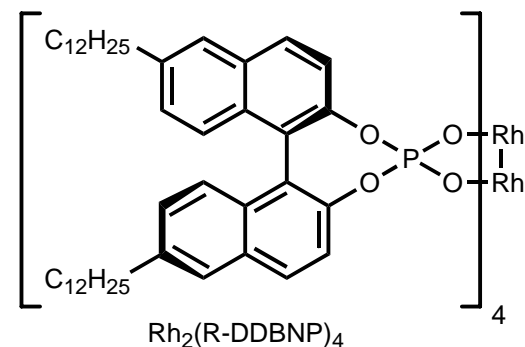
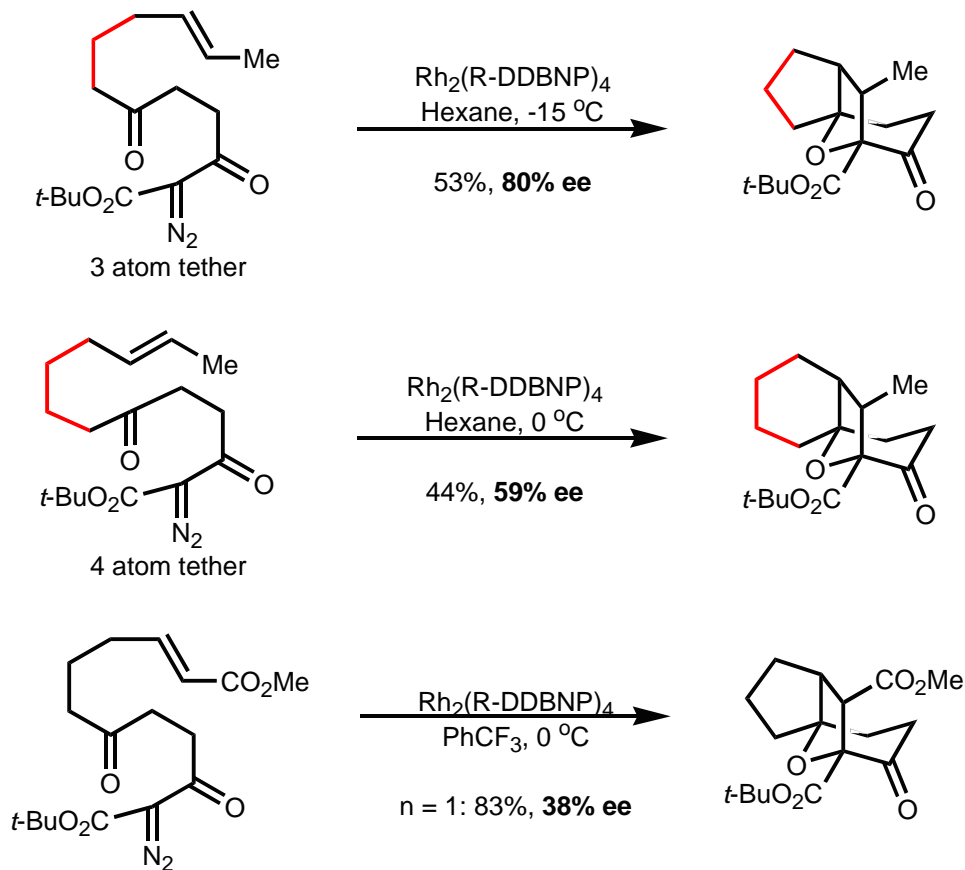
Doyle, M. P.; Forbes, D. C.; Protopopova, M. N.; Stanley, S. A.; Vasbinder, M. M.; Xavier, K.
R. J. Org. Chem. **1997**, 62, 7210.

Asymmetric Catalysis of Intramolecular [3 + 2] Cycloadditions



A highly reactive dienophile is required so that the cycloaddition occurs faster than metal decomplexation

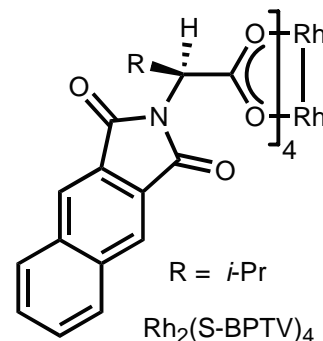
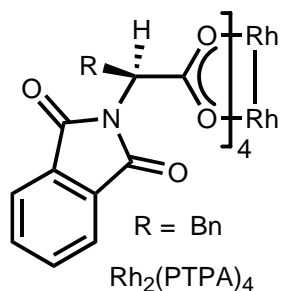
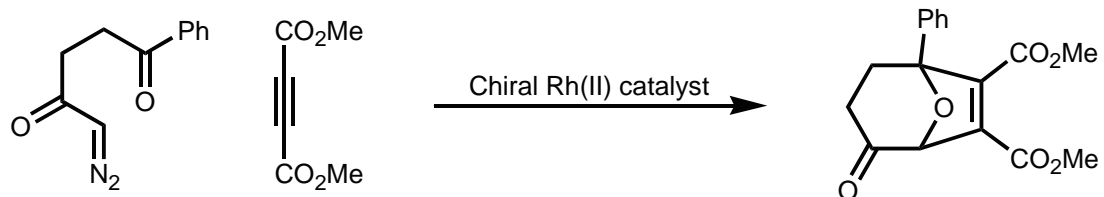
Substituted Alkene Tethers



The rate of the reaction is slower for the 4 atom tether. Therefore, the catalyst has more time to decomplex and form a free carbonyl ylide, resulting in lower ee's.

Hodgson, D. M.; Labande, A. H.; Pierard, F.; Expósito Castro, M. A. *J. Org. Chem.* **2003**, *68*, 6153.

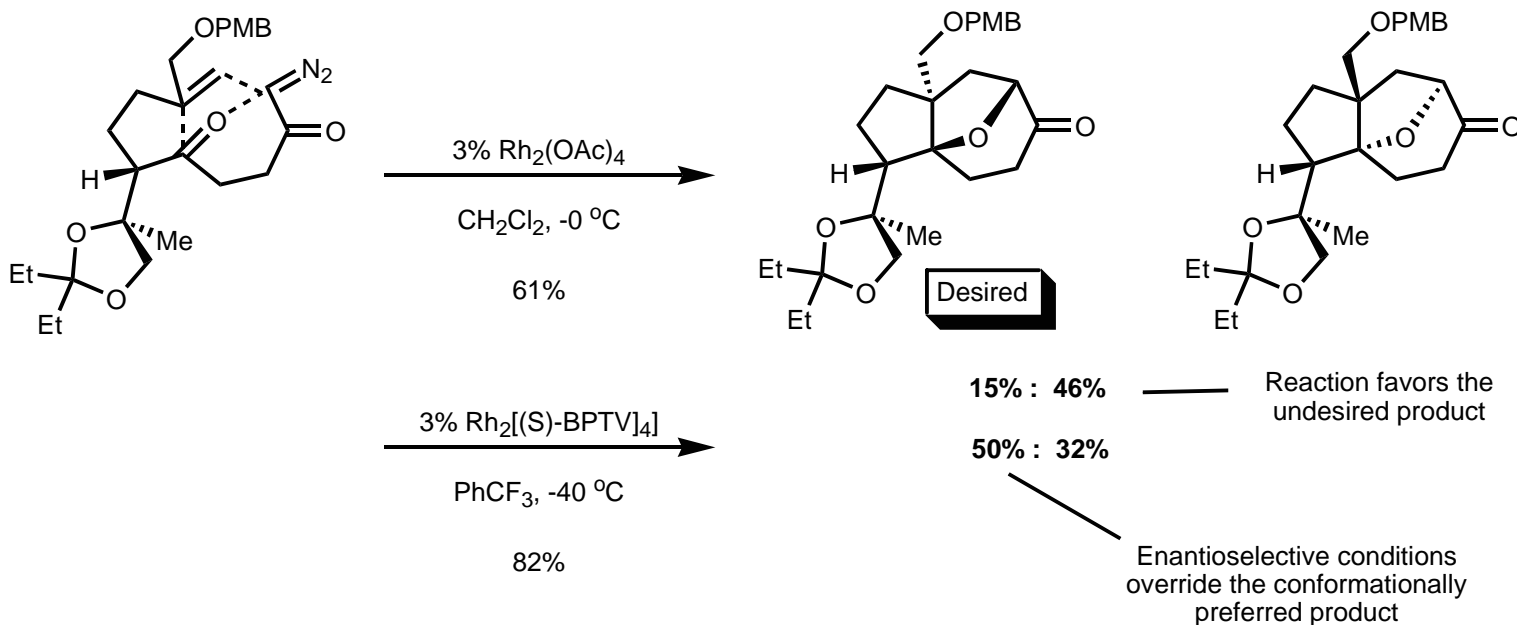
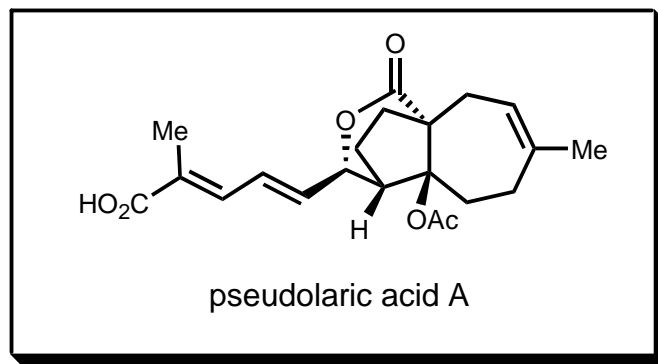
Asymmetric Catalysis of Intermolecular [3 + 2] Cycloaddition



Rh(II) catalyst	Solvent	Yield (%)	% ee
Rh ₂ (PTPA) ₄	CF ₃ C ₆ H ₅	81	60
Rh ₂ (PTPA) ₄	C ₆ H ₆	64	59
Rh ₂ (PTPA) ₄	CH ₂ Cl ₂	79	20
Rh ₂ (PTV) ₄	CF ₃ C ₆ H ₅	79	59
Rh ₂ (BPTV) ₄	CF ₃ C ₆ H ₅	79	90

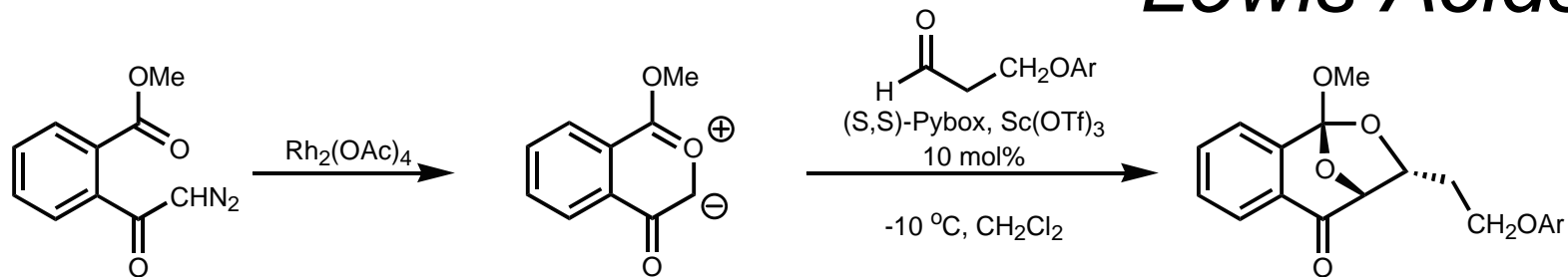
Kitagaki, S.; Yasugahira, M.; Anada, M.; Nakajima, M.; Hashimoto, S. *Tetrahedron Lett.* **2000**, *41*, 5931.

Enantioselective [3 + 2] Cycloaddition Cascade in Total Synthesis



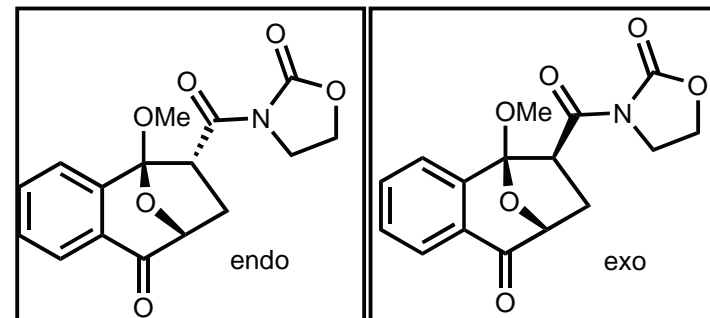
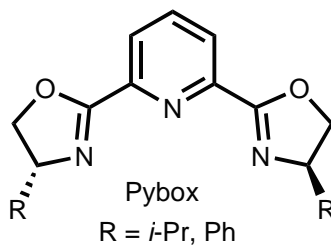
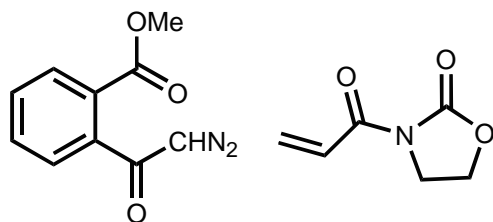
Geng, Z.; Chen, B.; Chiu, P. *Angew. Chem. Int. Ed.* **2006**, *45*, 6197.

Enantioselectivity Using Chiral Lewis Acids



77 - 97%,
70 - 90% endo, 82 - 93% ee

Ar: *p*-OMeC₆H₅, *o*-OMeC₆H₅,
p-FC₆H₅, *p*-ClC₆H₅, *p*-BrC₆H₅



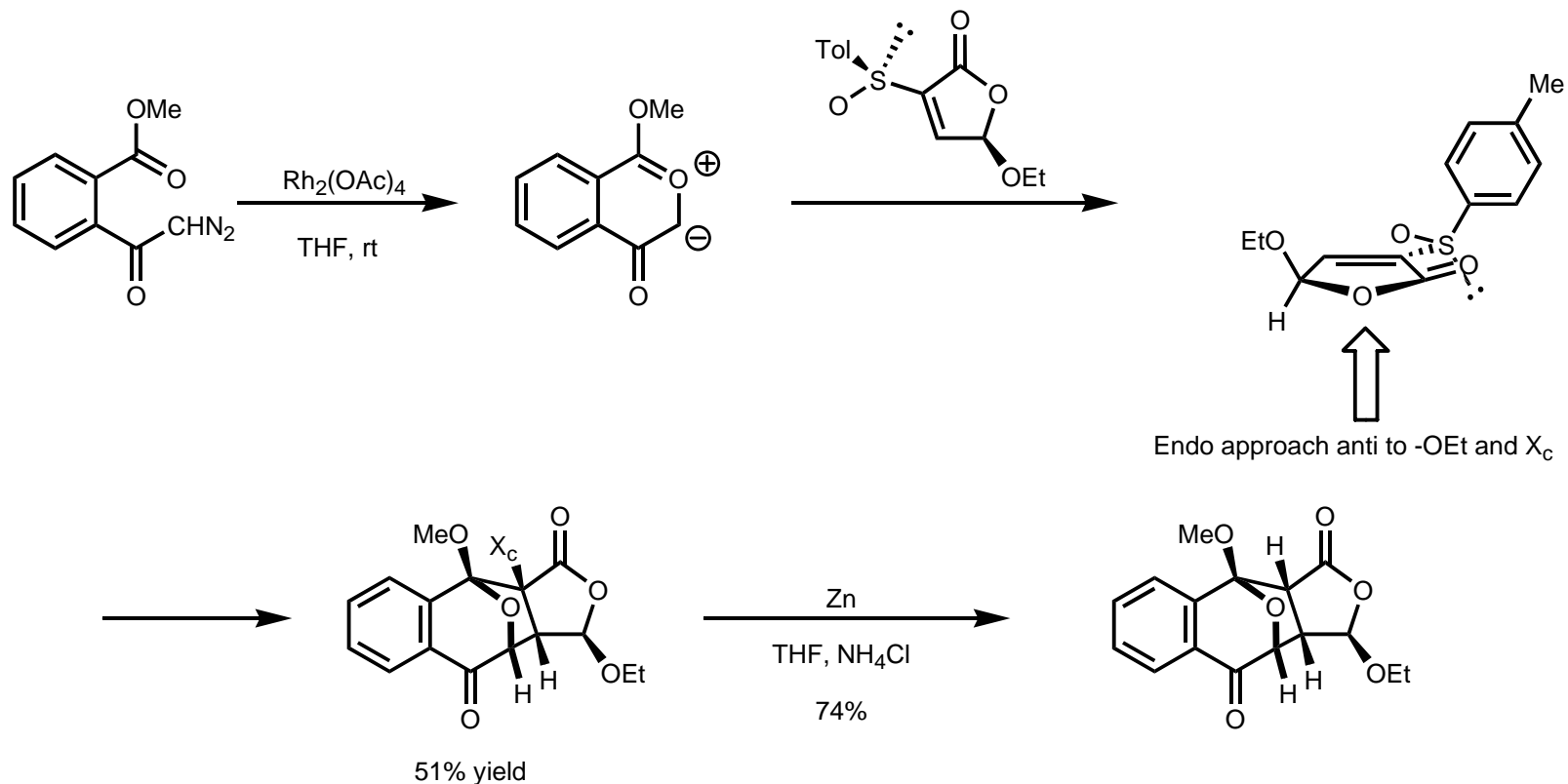
M = Sc: 89 : 11

endo adduct is favored
86%, 7% ee

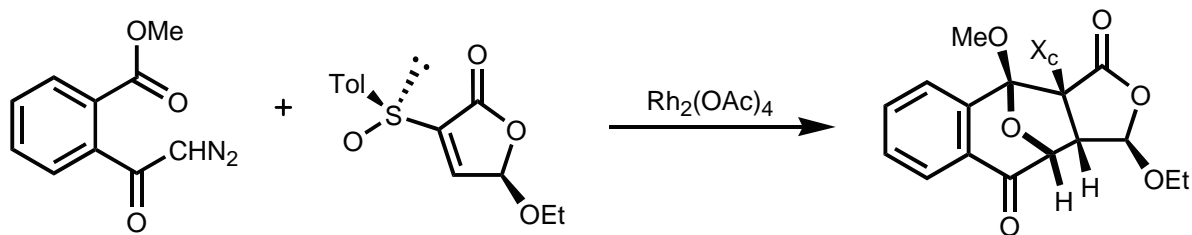
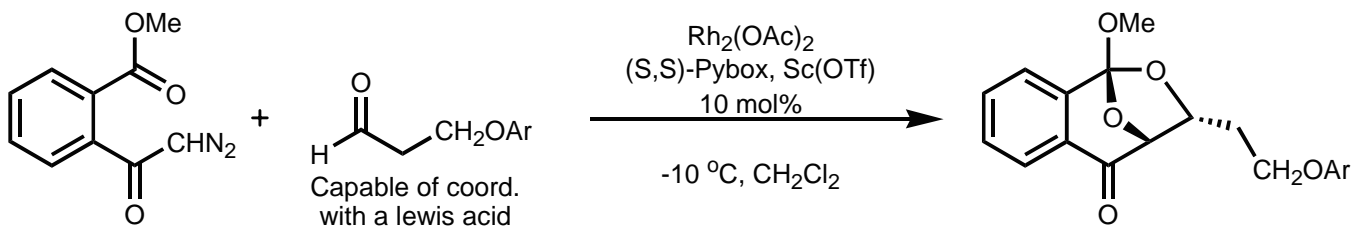
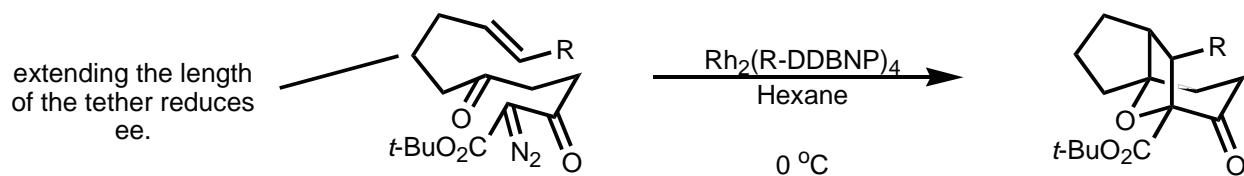
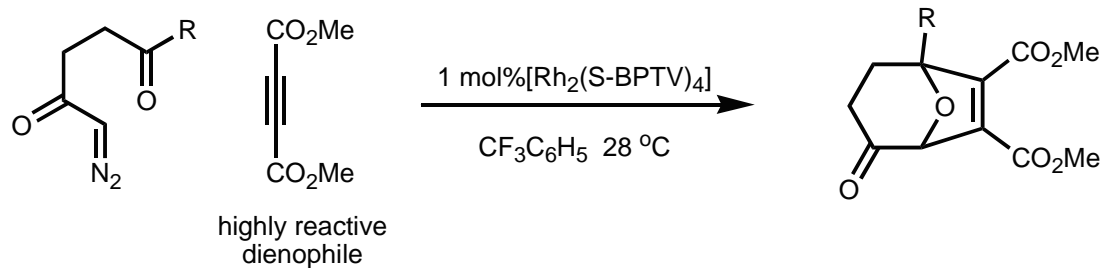
M = Yb: 12 : 88

exo adduct is favored
89%, 91% ee

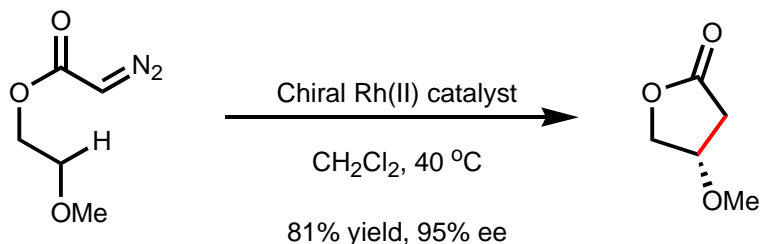
An Effective Chiral Auxiliary for [3 + 2] Cycloadditions



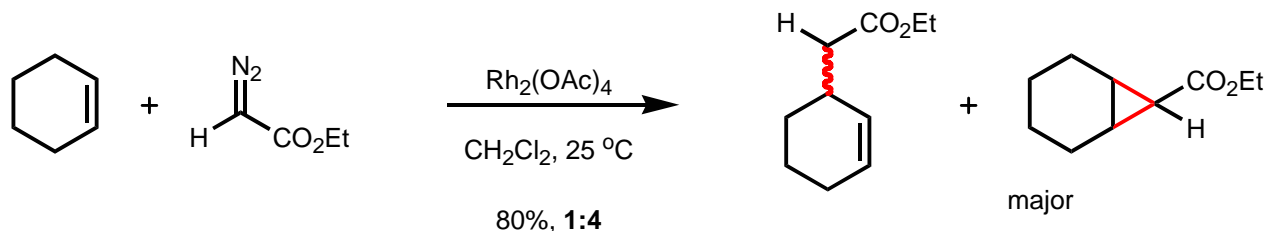
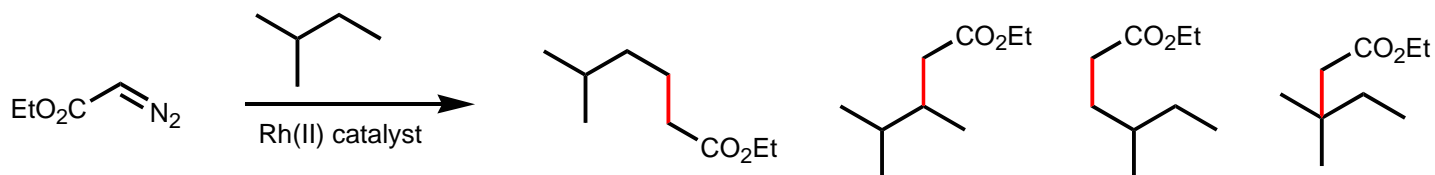
Summary of Enantioselective [3 + 2] Cycloaddition Cascade's



C-H Activation/Insertion



5 membered ring formation is preferred for intramolecular C-H activation

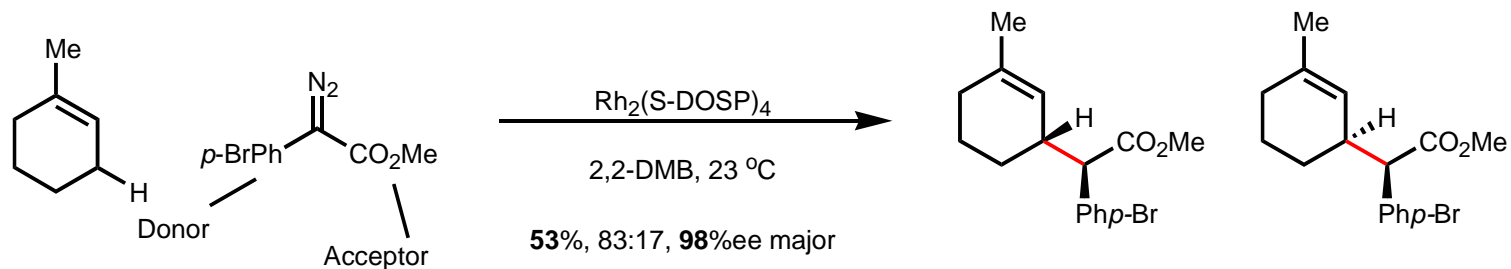


Cyclopropanation is preferred over C-H Activation/Insertion

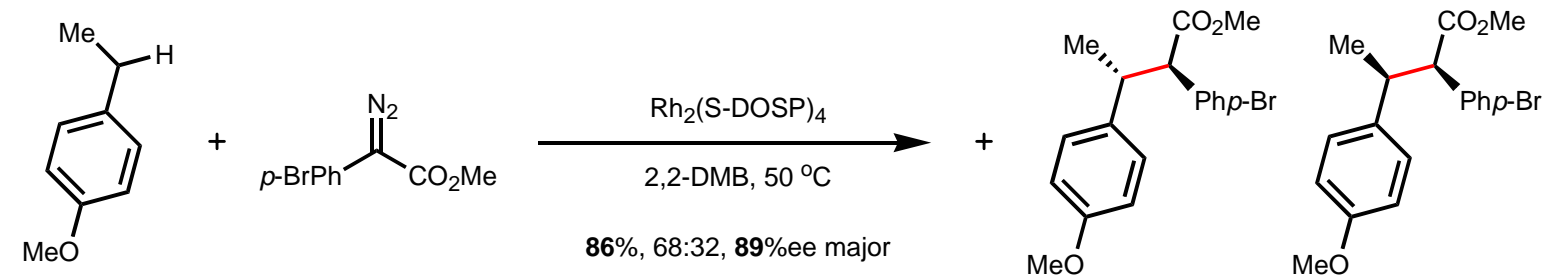
Regio- and chemoselectivity for the intermolecular reaction can be difficult

Davies, H. M. L.; Beckwith, R. E. J. *Chem. Rev.* **2003**, 103, 2861.

C-H Activation/Insertion



Allylic Activation

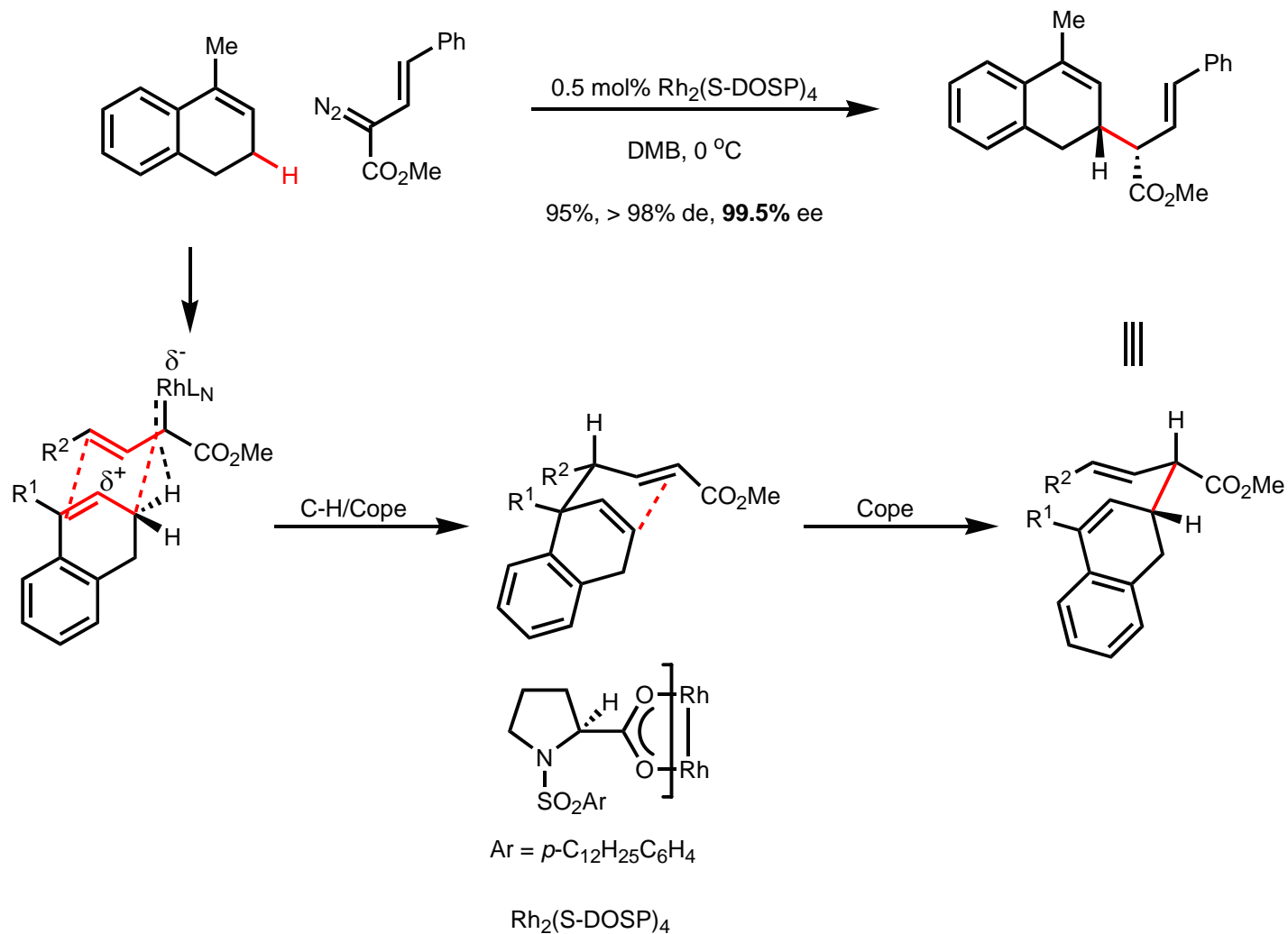


Electron Rich
Substrate

Secondary Benzylic
C-H Activation

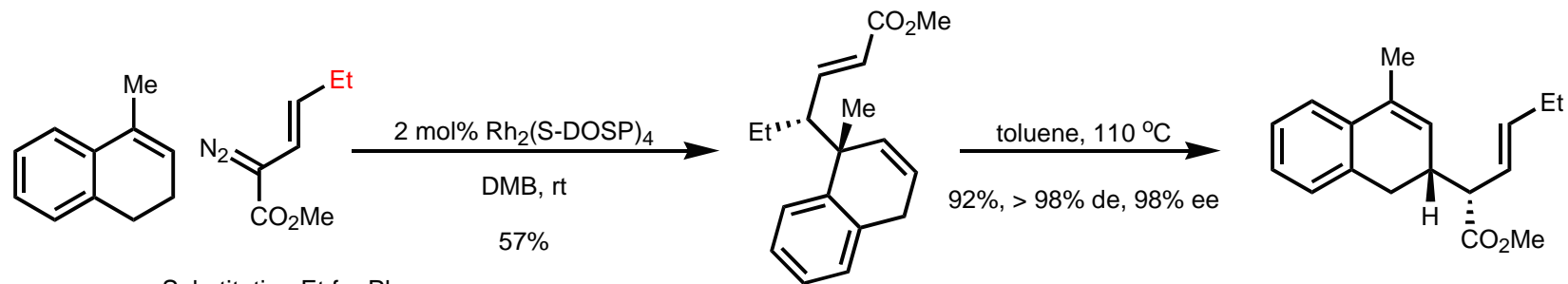
Donor/Acceptor carbenoids paired with electron rich substrates allows for intermolecular C-H activation with diminished cyclopropane byproducts

C-H Activation/Cope Rearrangement

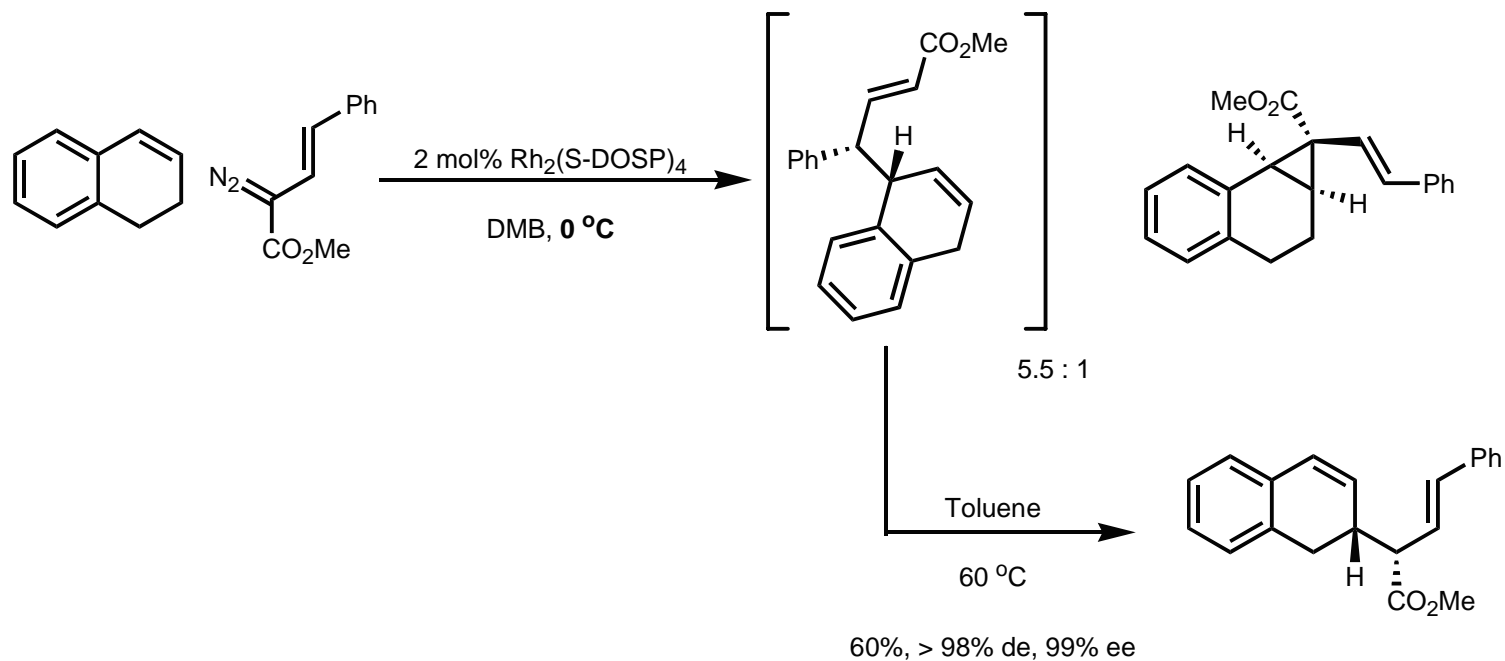


Davies, H. M. L.; Jin, Q. *J. Am. Chem. Soc.* **2004**, *126*, 10862.

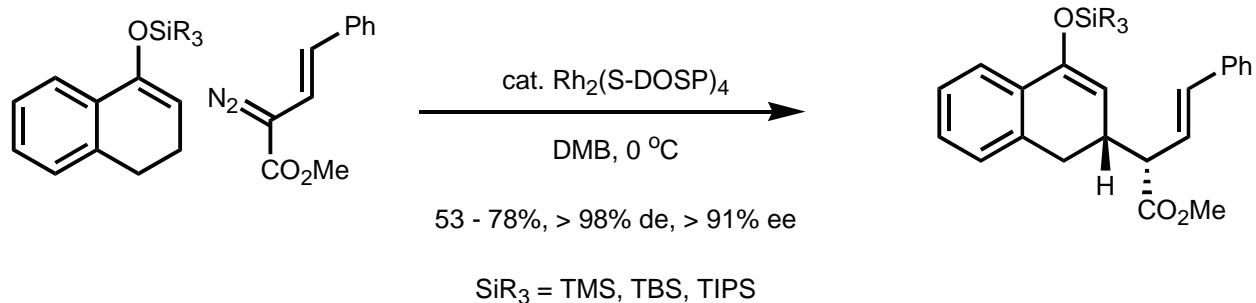
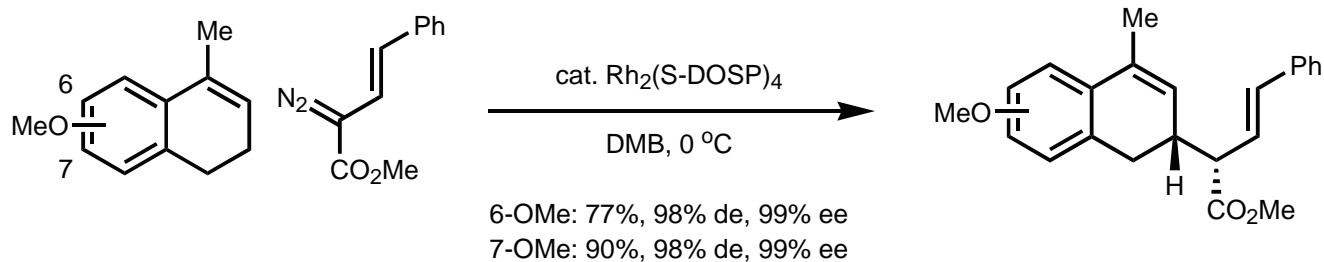
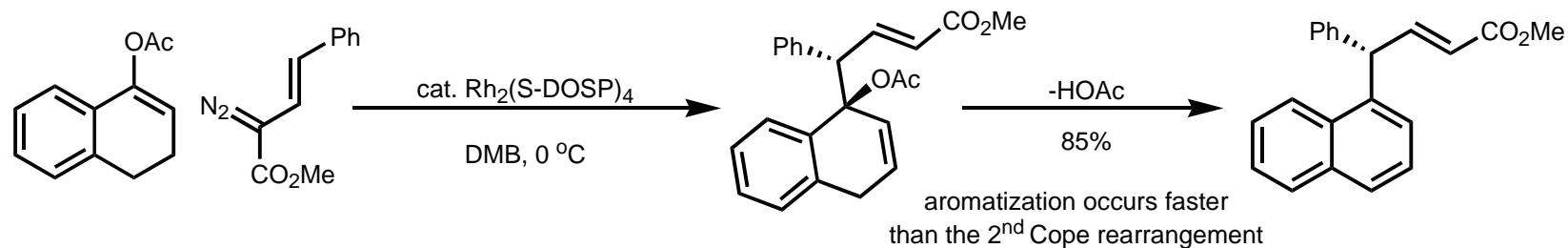
Support for the Proposed Mechanism



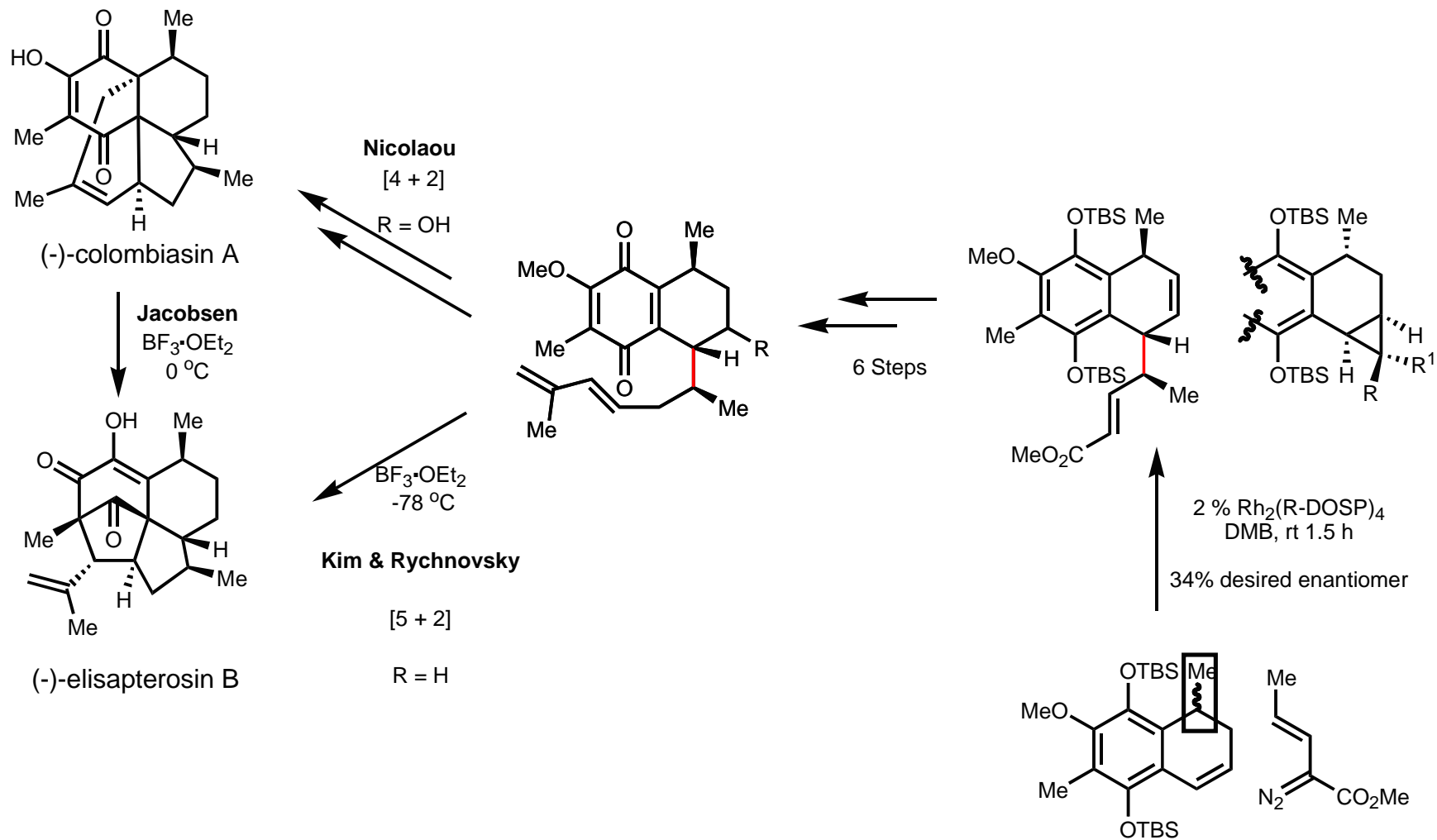
Substituting Et for Ph
attenuates the reactivity



Applicable Substrates

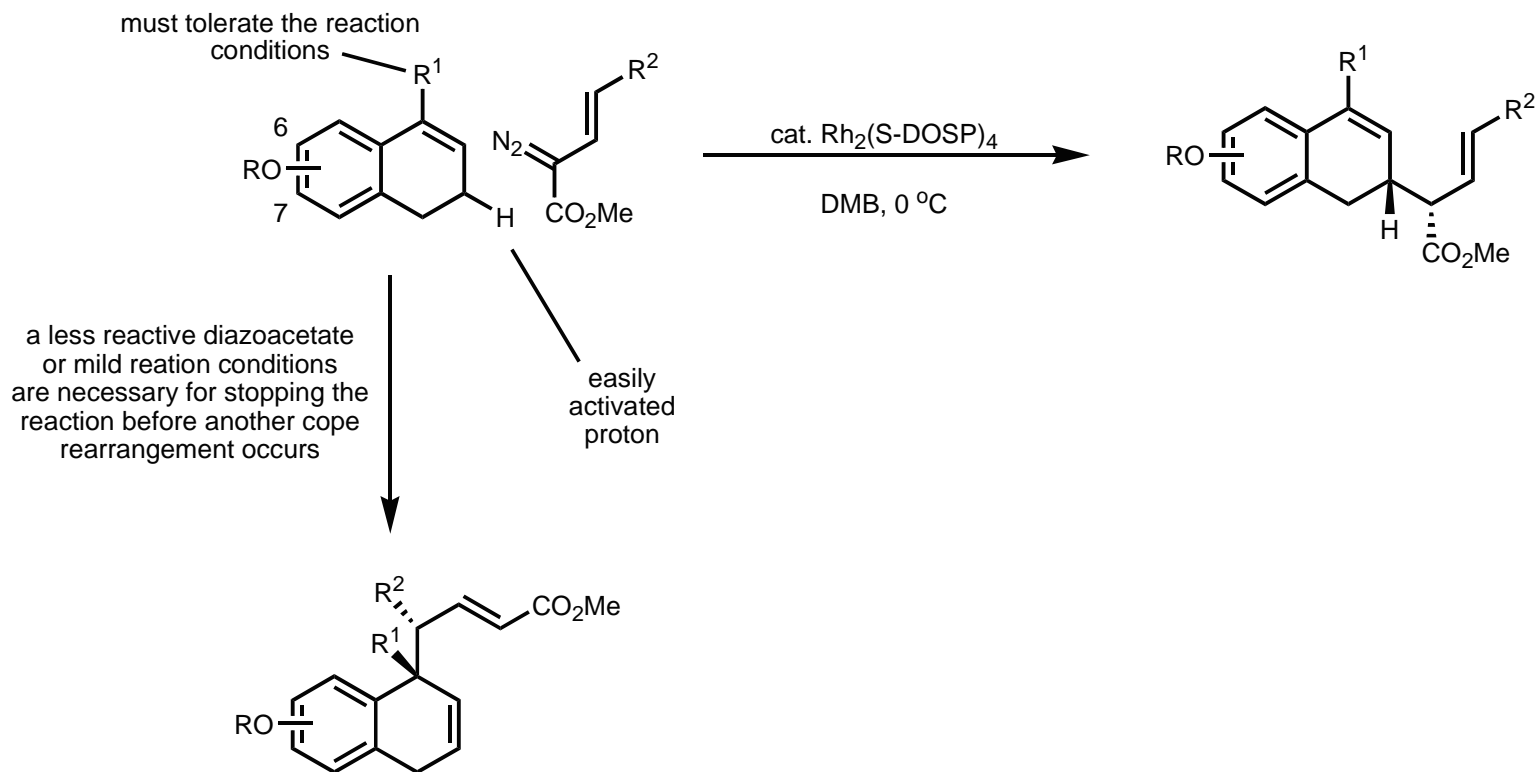


Total Synthesis of (-)-Colombiasin A and (-)-Elisapterosin B

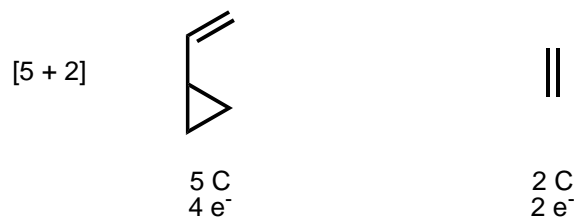
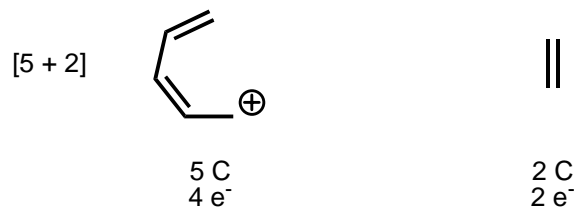
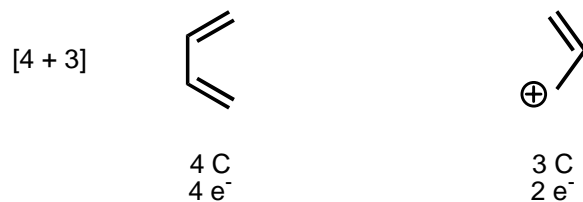


Davies, H. M. L.; Dai, X.; Long, M. S. *J. Am. Chem. Soc.* **2005**, *128*, 2485.

Summary of Enantioselective C-H Activation/Cope Cascades

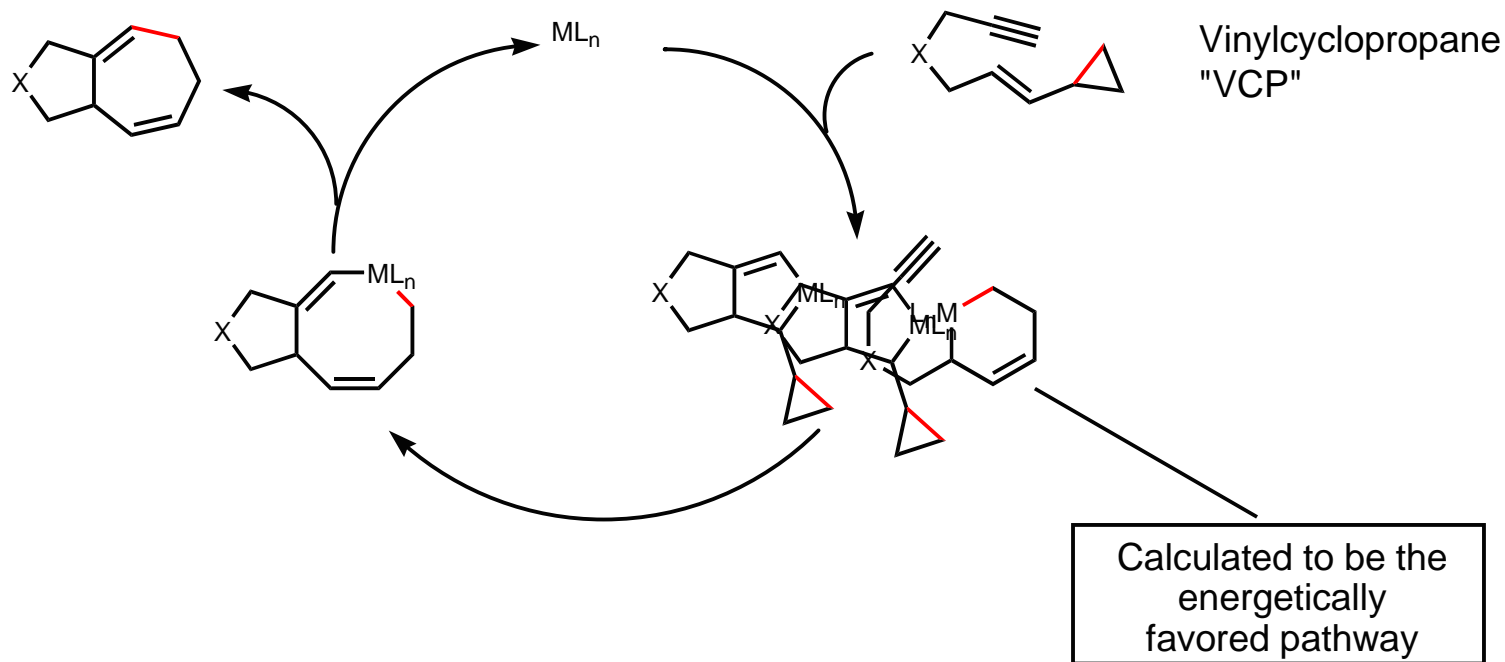


[5 + 2] Cycloadditions

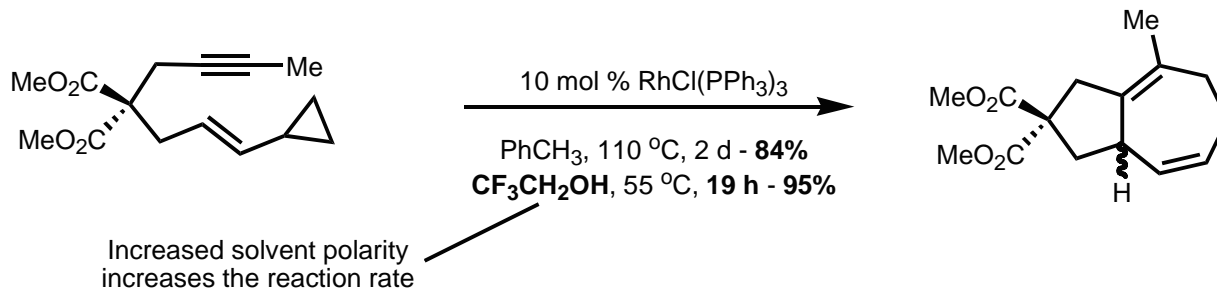


Wender, P. A.; Husfeld, C. O.; Langkopf, E.; Love, J. A.; Pleuss, N.
Tetrahedron **1998**, 54, 7203.

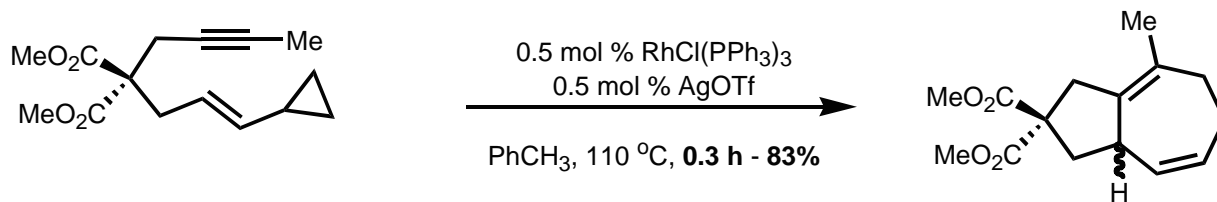
Concept



Reaction Development



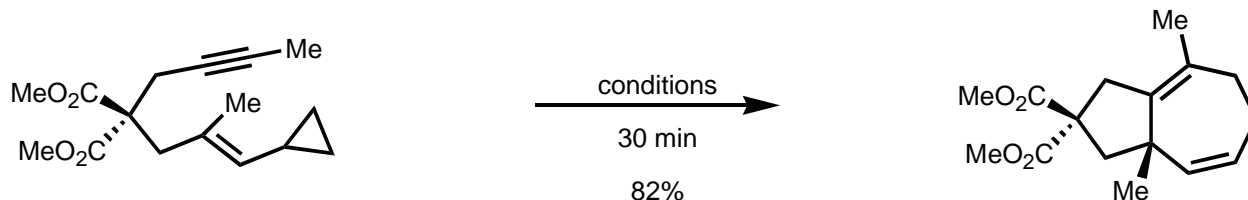
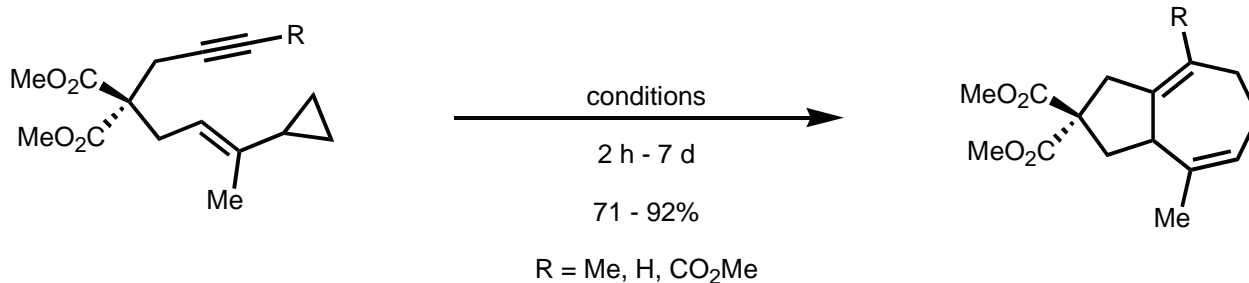
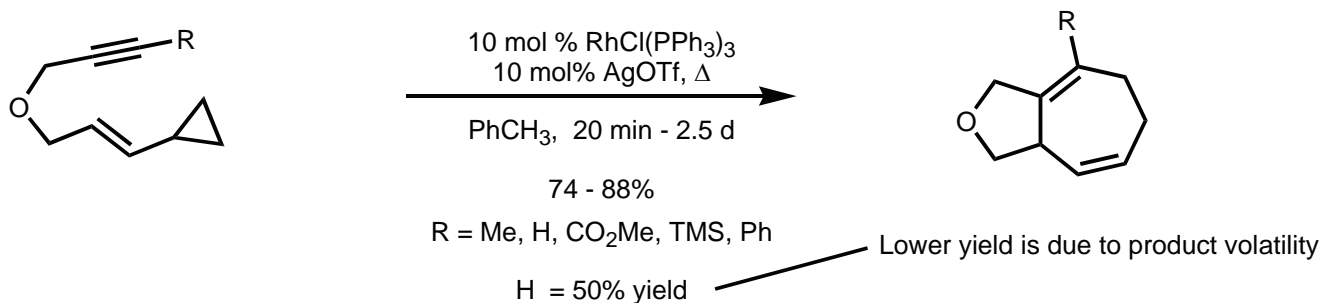
The increased solvent polarity is believed to favor formation of a cationic rhodium species through loss of chloride ligand



The silver additive irreversibly forms a cationic rhodium(I) species, opening up a vacant coordination site on the metal, through removal of the chloride ligand

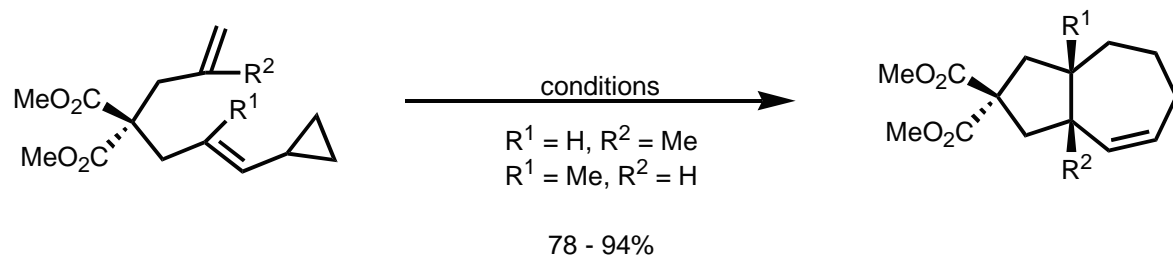
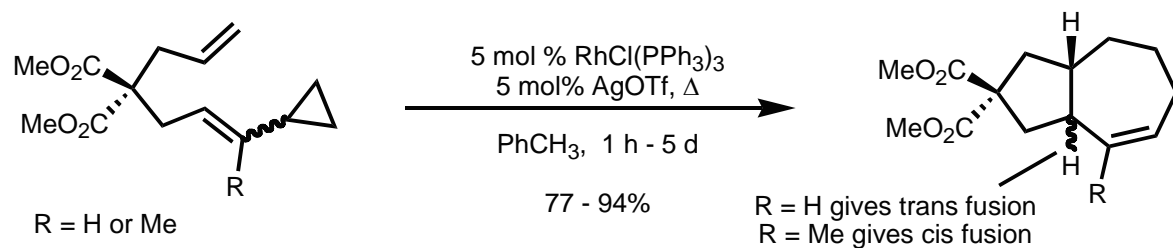
Wender, P. A.; Takahashi, H.; Witulski, B. *J. Am. Chem. Soc.* **1995**, 117, 4720. Wender, P. A.; Husfeld, C. O.; Langkopf, E.; Love, J. A.; Pleuss, N. *Tetrahedron* **1998**, 54, 7203.

Intramolecular [5 + 2] Cycloadditions VCP's and Alkynes



Wender, P. A.; Takahashi, H.; Witulski, B. *J. Am. Chem. Soc.* **1995**, 117, 4720. Wender, P. A.; Husfeld, C. O.; Langkopf, E.; Love, J. A.; Pleuss, N. *Tetrahedron* **1998**, 54, 7203.

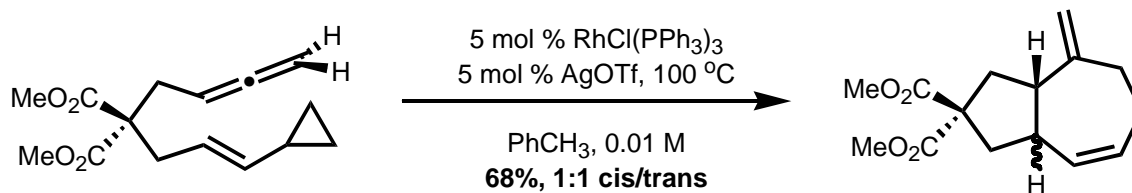
Intramolecular [5 + 2] Cycloadditions VCP's and Alkenes



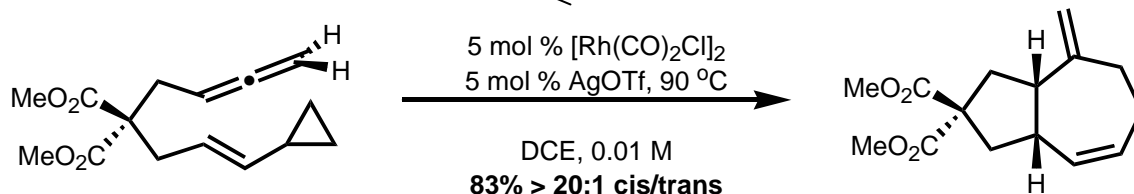
A terminal olefin is required as substituted alkenes produce mainly β -elimination products

Wender, P. A.; Husfeld, C. O.; Langkopf, E.; Love, J. A.; *J. Am. Chem. Soc.* **1998**, *120*, 1940.
Wender, P. A.; Husfeld, C. O.; Langkopf, E.; Love, J. A.; Pleuss, N. *Tetrahedron* **1998**, *54*, 7203.

New Selective Catalyst



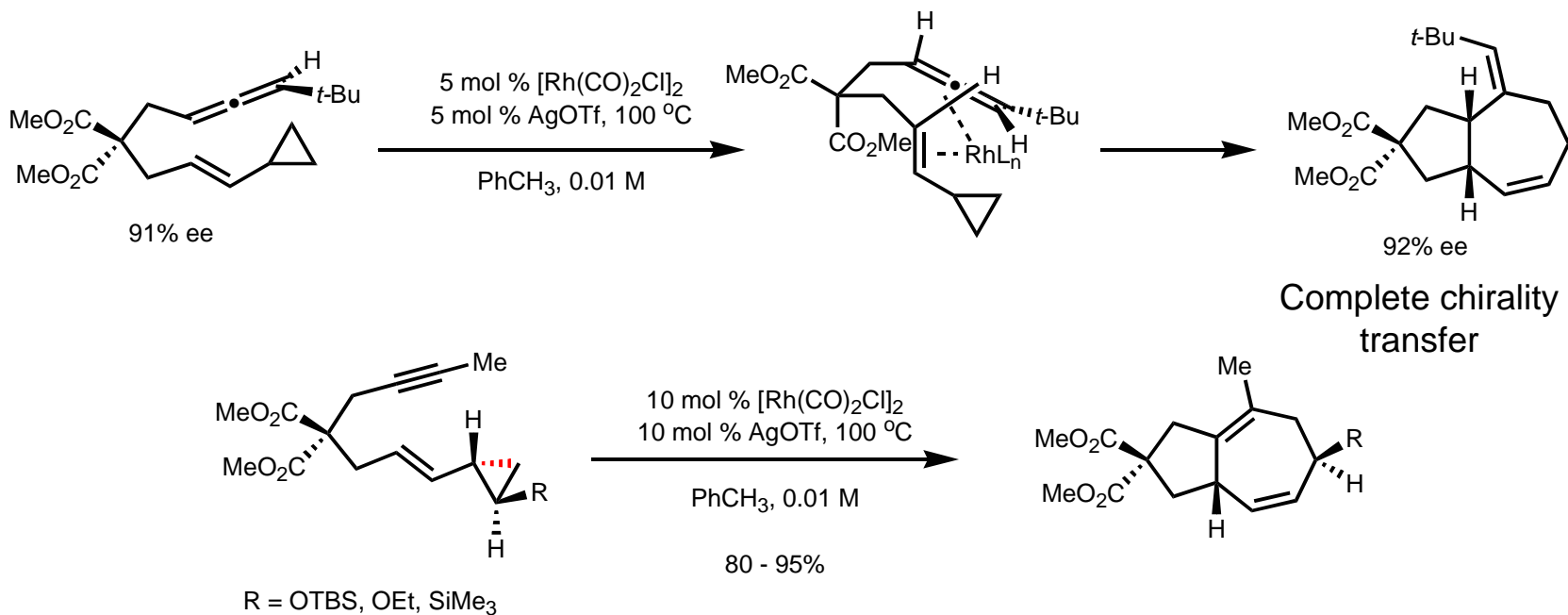
Dimeric Rh(I) catalyst
provides increased
rxn rates



The active catalyst, presumed to be Rh(CO)₂Cl, is thought to be more active as the lower ligand count facilitates coordination to the metal

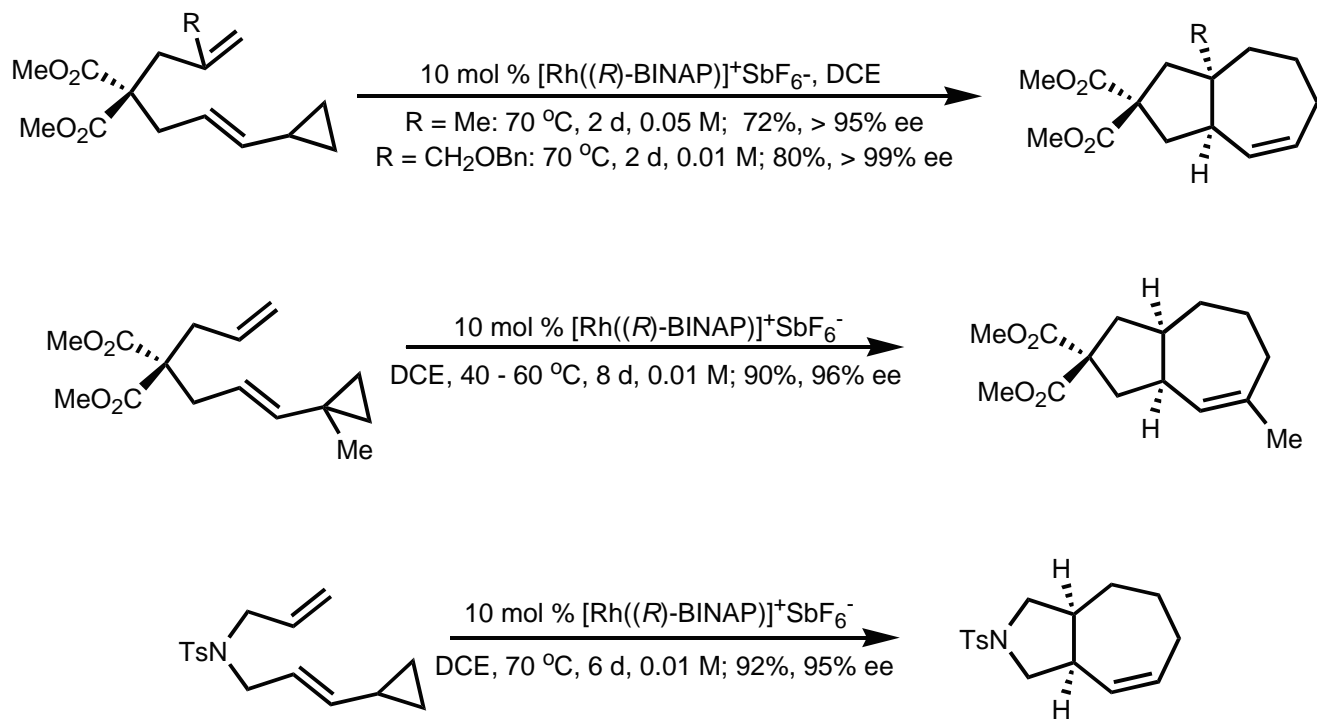
Wender, P. A.; Rieck, H.; Fuji, M. *J. Am. Chem. Soc.* **1998**, *120*, 10976. Love, J. A.; *J. Am. Chem. Soc.* **1999**, *121*, 5348.

Stereochemical Information



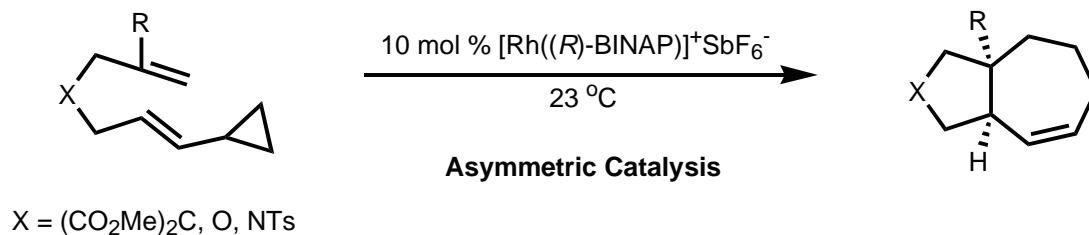
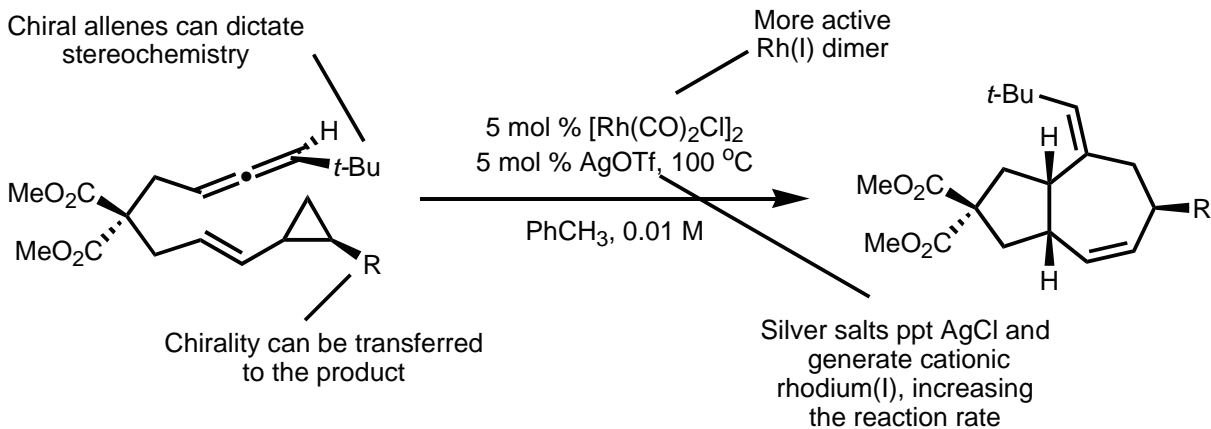
Wender, P. A.; Glorius, F.; Husfeld, C. O.; Langkopf, E.; Love, J. A.; *J. Am. Chem. Soc.* **1999**, *121*, 5348. Wender, P. A.; Dyckman, A. J.; Husfeld, C. O.; Kadereit, D.; Love, J. A.; Reick, H. *J. Am. Chem. Soc.* **1999**, *121*, 10442.

Asymmetric Catalysis of the [5 + 2] Cycloaddition

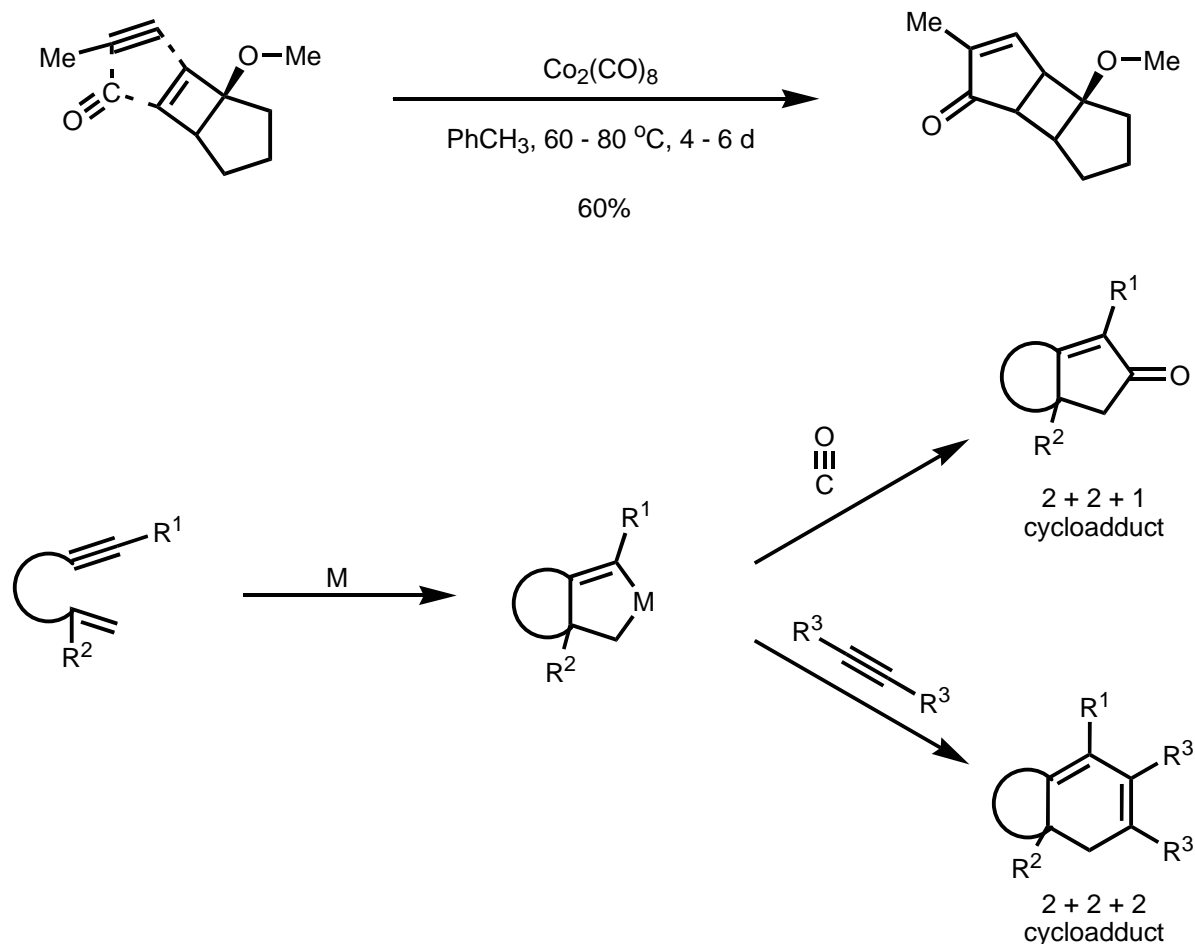


Wender, P. A.; Haustedt, L. O.; Lim, J.; Love, J. A.; Williams, T. J.; Yoon, J. *J. Am. Chem. Soc.* **2005**, *128*, 6302.

Summary of Enantioselective [5 + 2] Cycloaddition Cascades

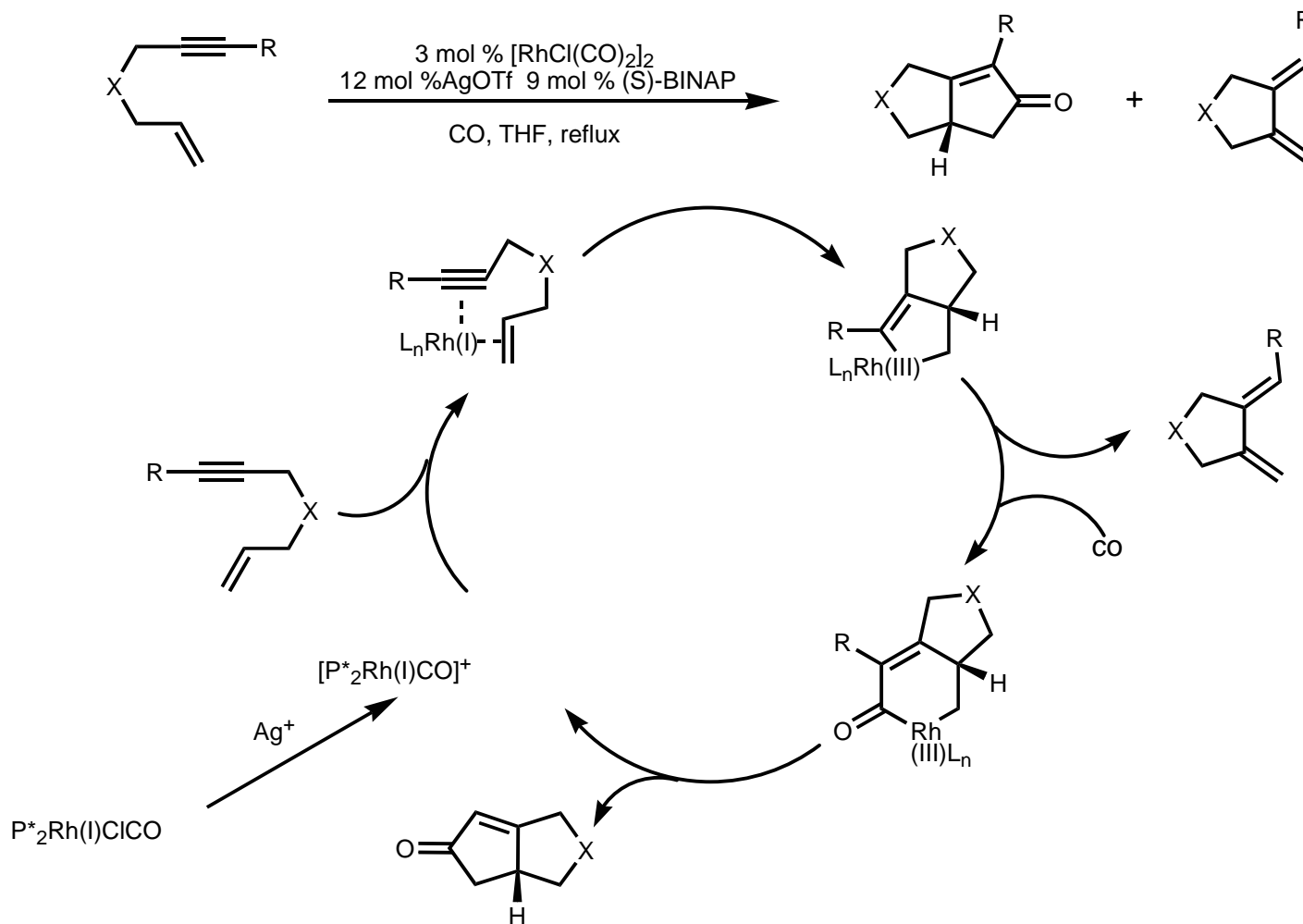


Three Component Cycloadditions



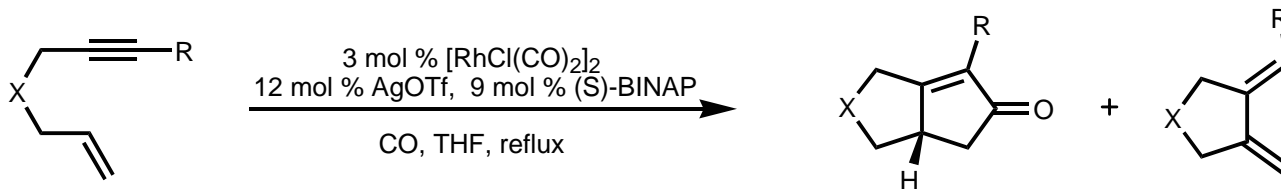
Pauson, P. L.; Khand, I. U. *Ann. N.Y. Acad. Sci.* **1977**, 295, 2. Pauson, P. L.; *Tetrahedron*, **1985**, 41, 5855. Schore, N. E.; *Chem. Rev.* **1988**, 88, 1081.

Rhodium Catalyzed Asymmetric PKR



Jeong, N.; Sung, B. K.; Choi, Y. K. *J. Am. Chem. Soc.* **2000**, 122, 6771.

Reaction Conditions

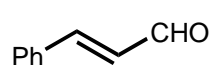
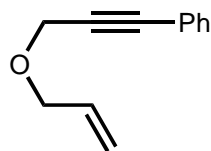
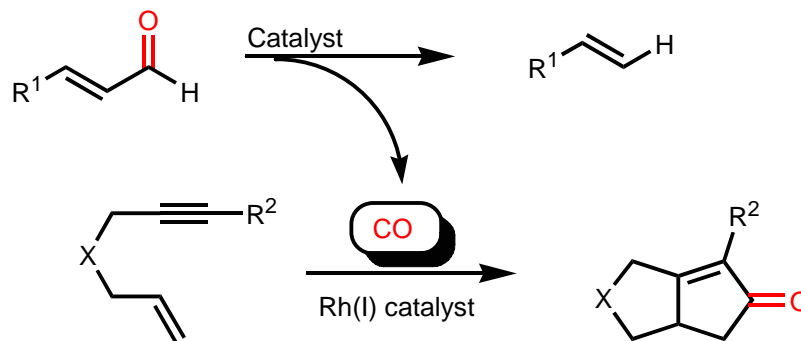


X	R	CO (atm)	temp (C)	time (h)	yield (%)	ee (%)
O	Me	2	130	20	85	86
O	Me	1	90	5	40	96
O	Ph	1	90	5	88	81

- High temperature and pressure increases yield and decreases ee
- Low temperature and pressure increases ee and decreases yield

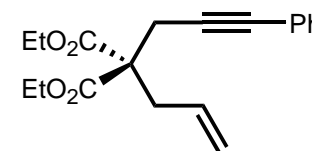
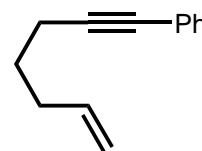
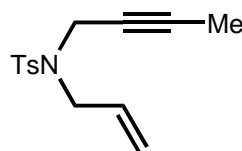
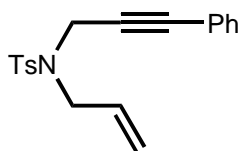
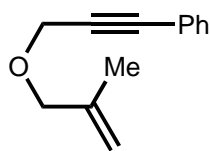
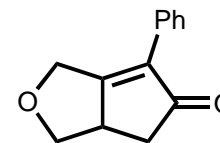
Jeong, N.; Sung, B. K.; Choi, Y. K. *J. Am. Chem. Soc.* **2000**, 122, 6771.

Rhodium Catalyzed Decarbonylation / APKR



5 mol % Rh(dppp)₂Cl

20 equiv CHO: 120 °C, 2h, No Solvent: **98%**
 5 equiv CHO: 120 °C, 3h, No Solvent: **93%**
 1.2 equiv CHO: 120 °C, 3h, No Solvent: **83%**

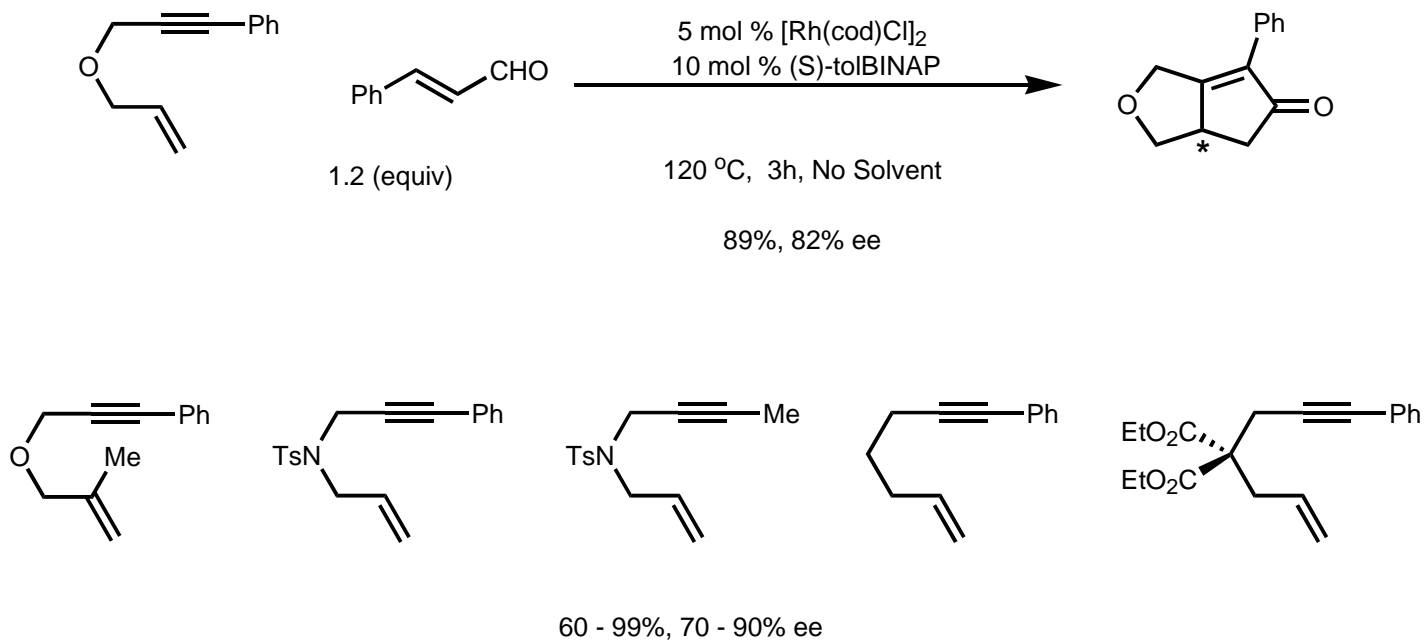


58 - 85% yield

- Aromatic and α , β -unsaturated aldehydes work best for decarbonylation

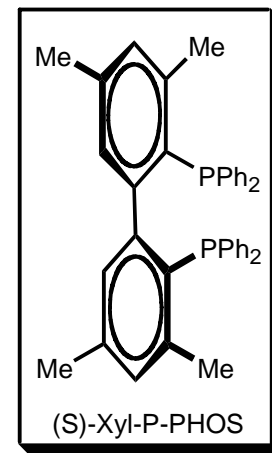
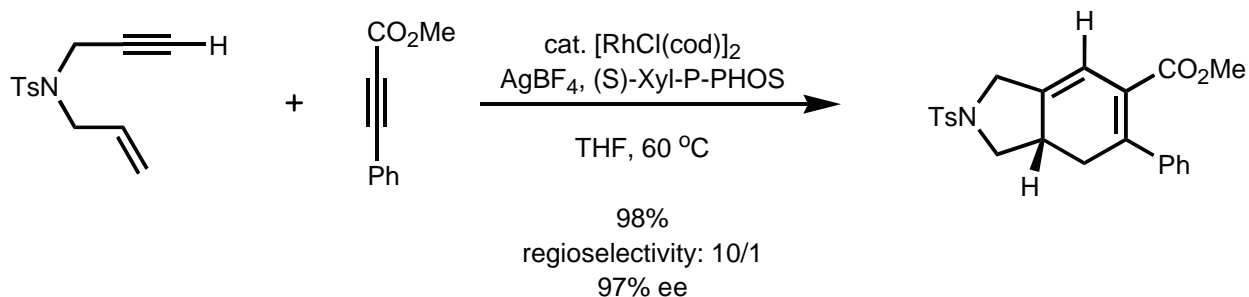
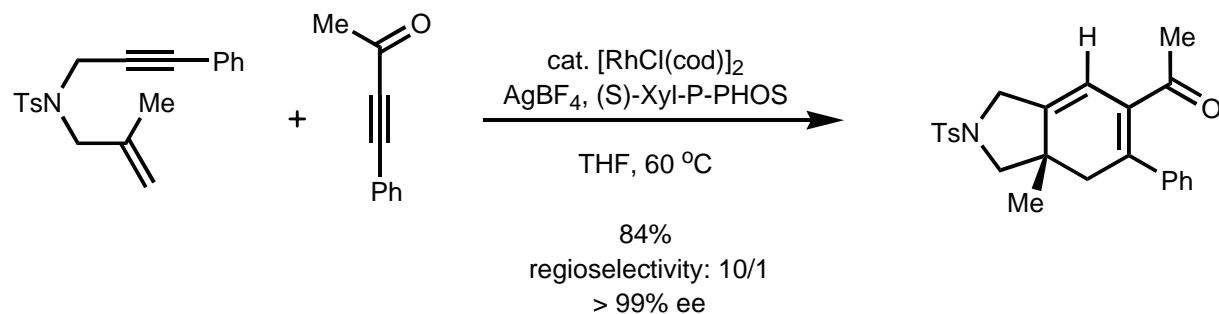
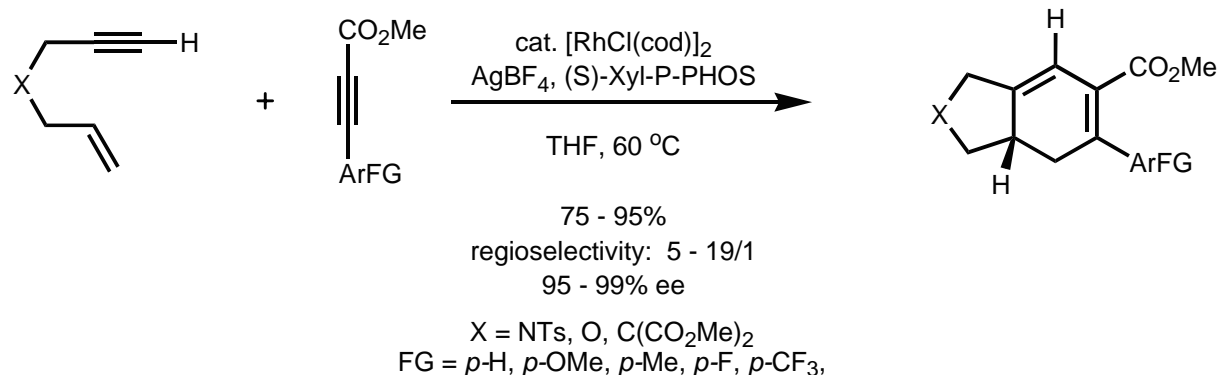
Shibata, T.; Toshida, N.; Takagi, K. *J. Org. Chem.* **2002**, 67, 7446.

Enantioselective Conditions

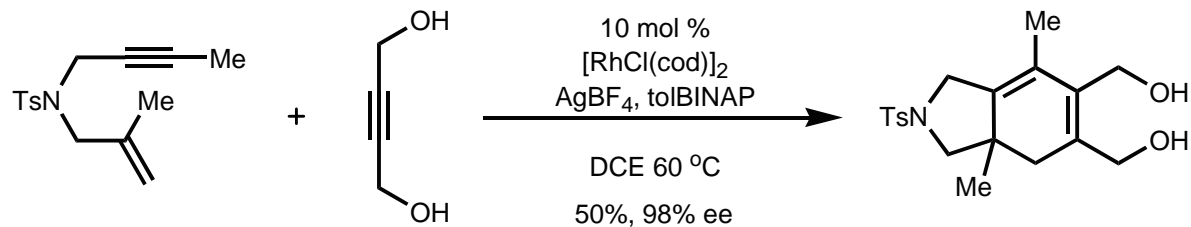
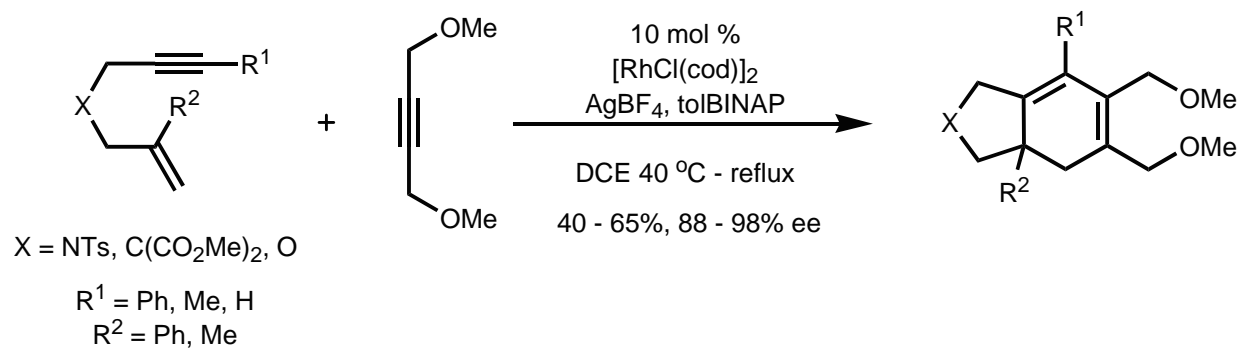


Shibata, T.; Toshida, N.; Takagi, K. *J. Org. Chem.* **2002**, 67, 7446.

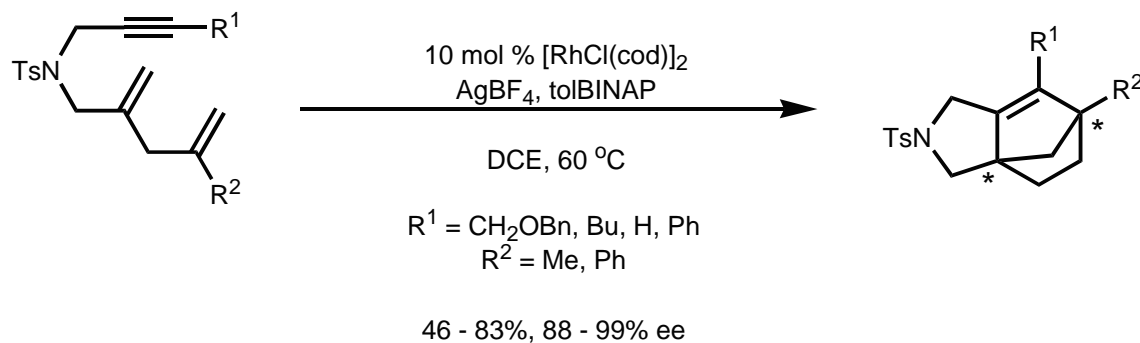
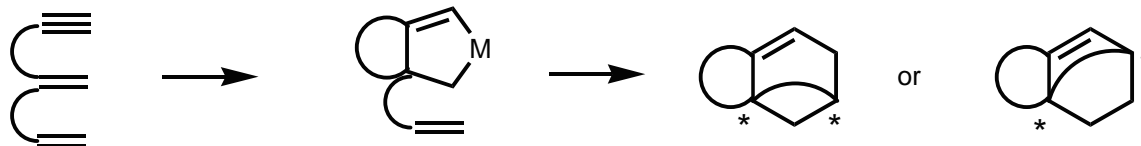
Regioselective [2 + 2 + 2] Cycloadditions



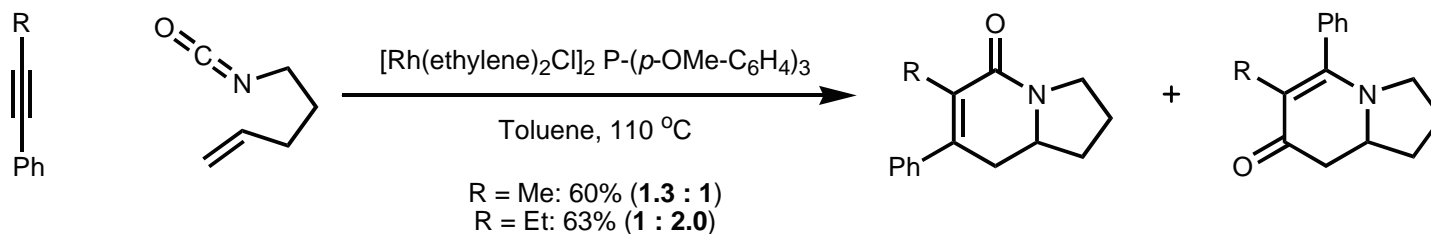
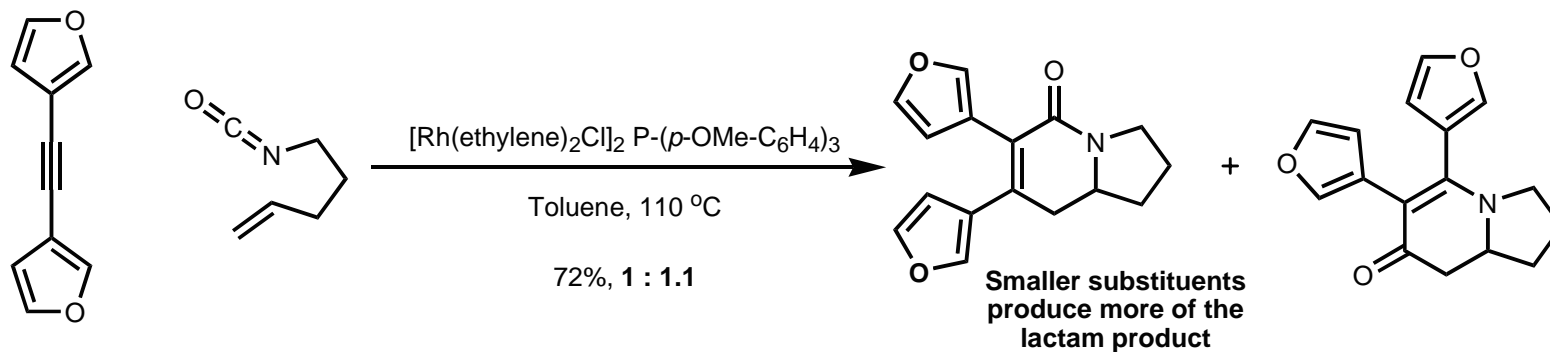
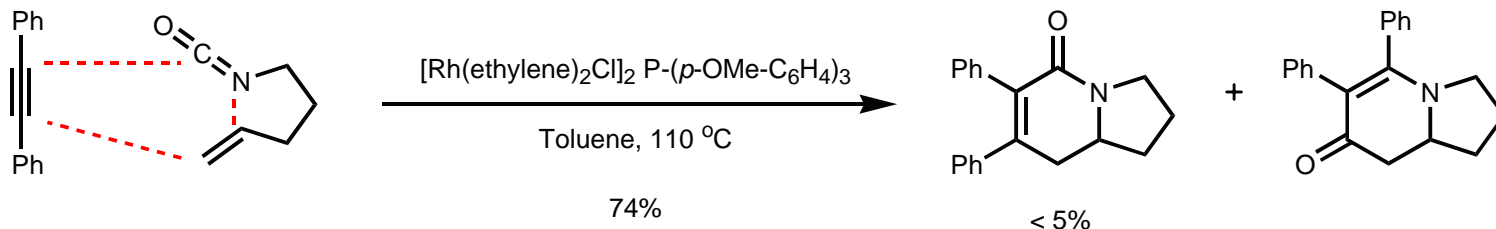
Reaction Conditions



Enantioselective Intramolecular [2 + 2 + 2]'s

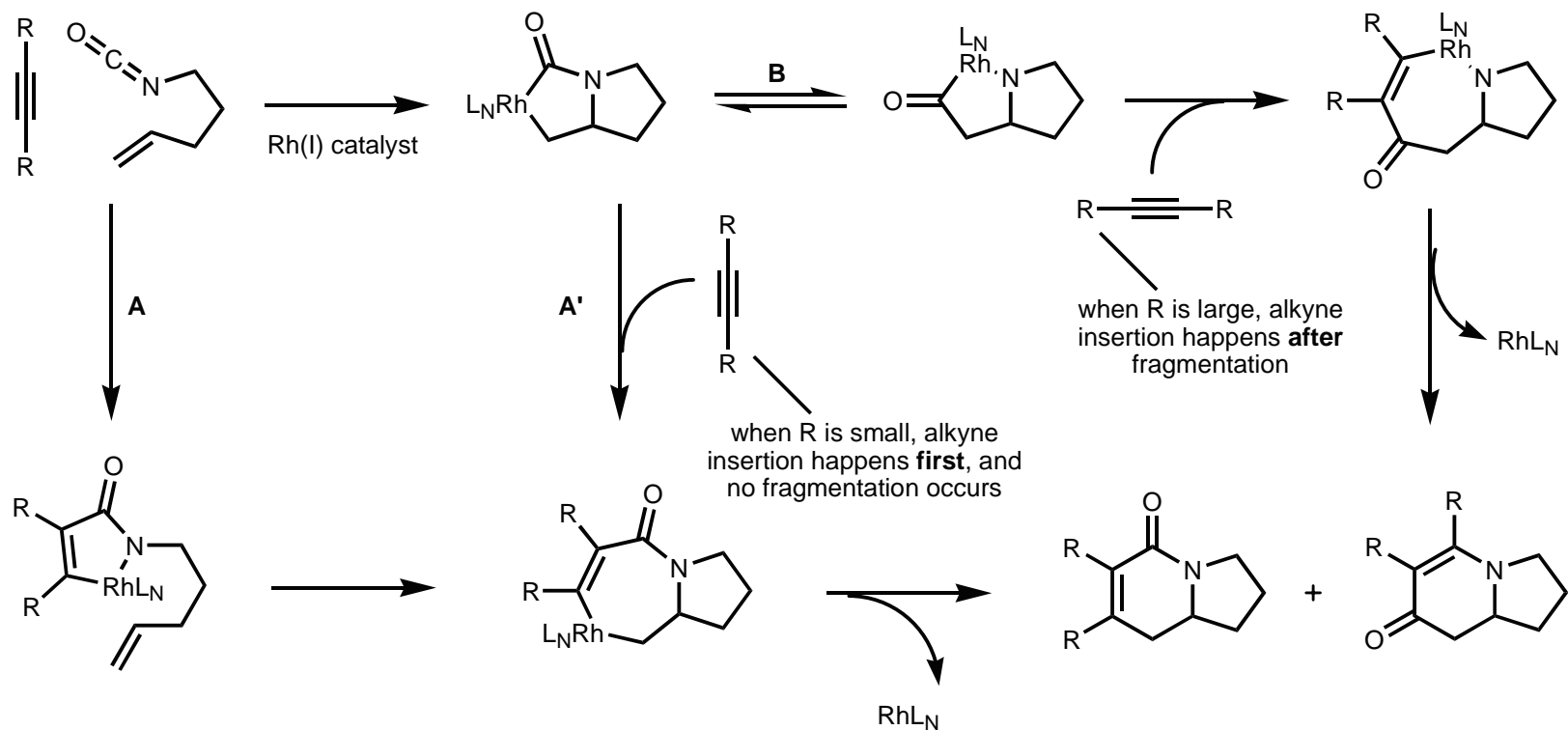


[2 + 2 + 2]'s with Isocyanates



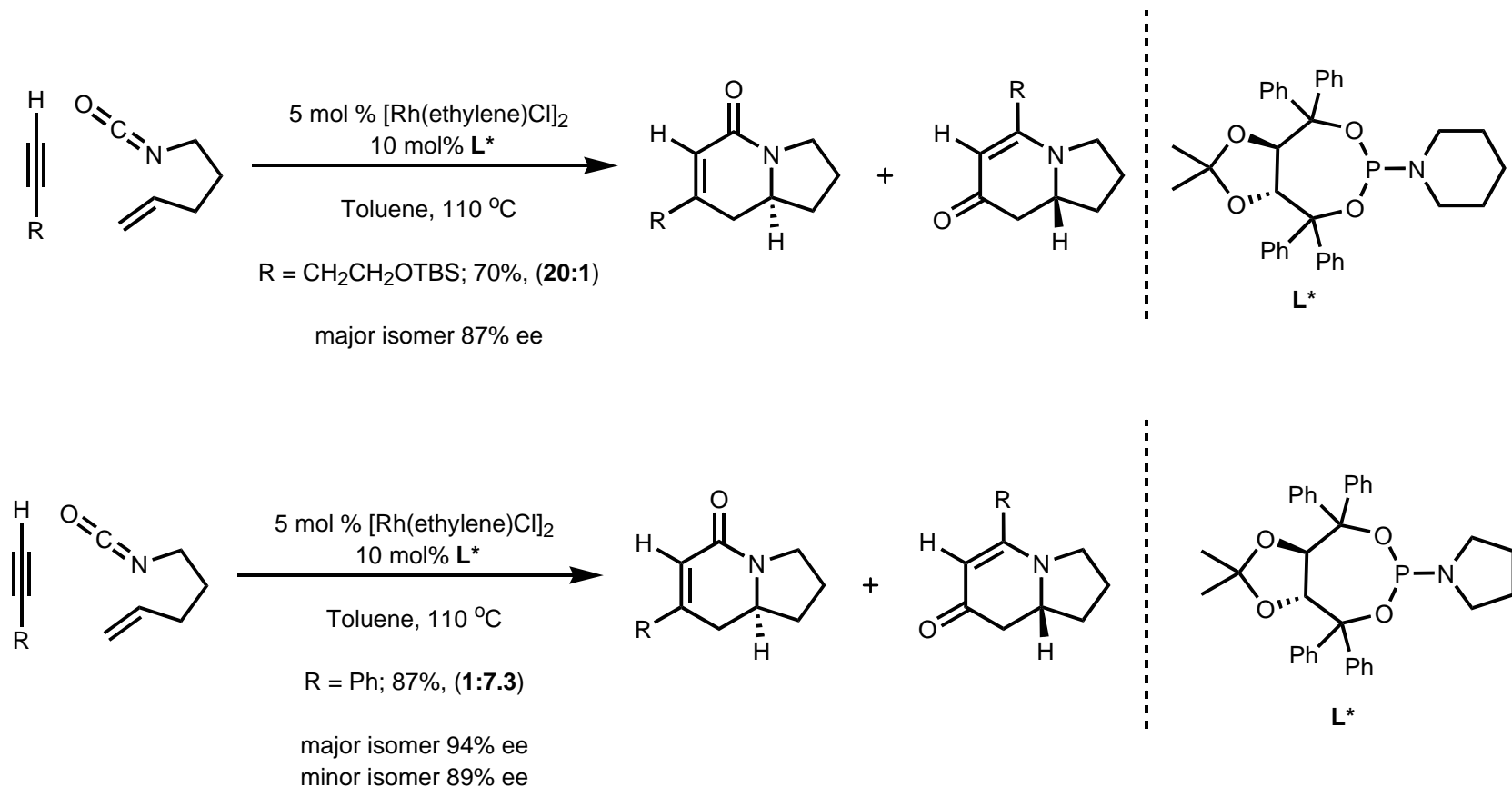
One extra carbon
reverses selectivity

Possible Mechanism

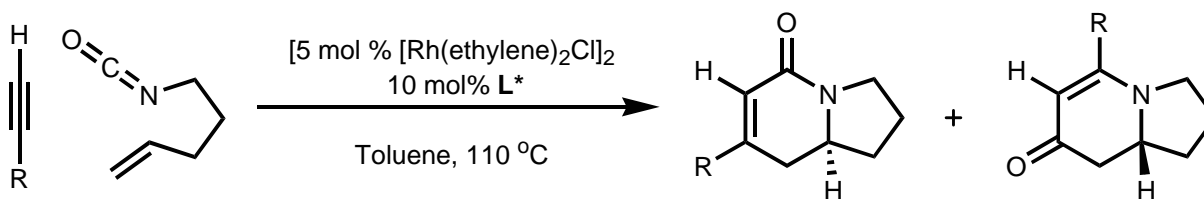
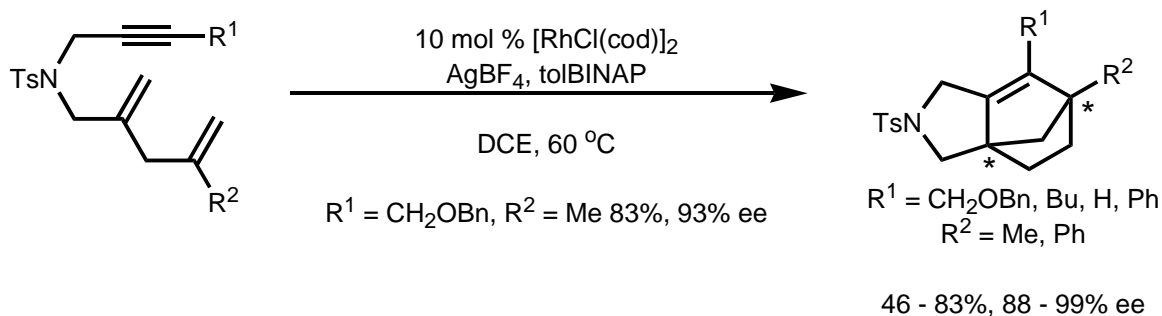
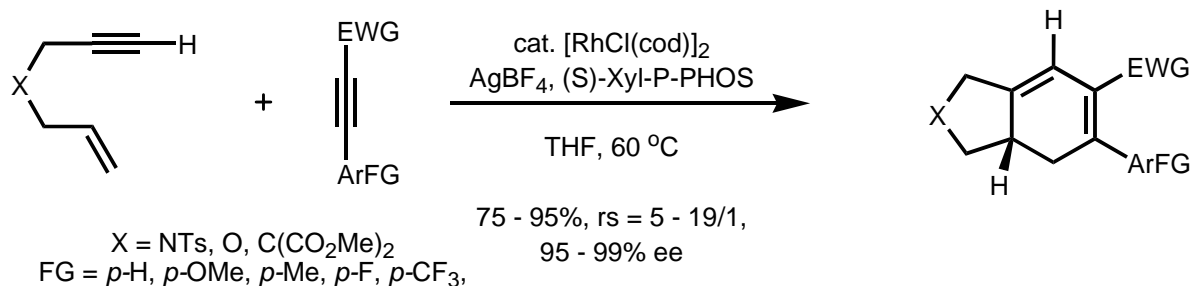


In pathway B CO migration is believed to occur faster than insertion of bulky alkynes. Thus smaller more reactive alkynes give products via pathway A'.

Asymmetric [2 + 2 + 2] Cycloaddition with Isocyanates



Summary of Enantioselective Three Component Cycloadditions



In Conclusion

- The unique reactivity of Rhodium has allowed for the realization of impressive reaction cascades that generate molecular complexity in a concise fashion.
- The continuous development of new chiral Rhodium catalysts has allowed for the development of efficient methods of asymmetric catalysis.
- Rhodium has shown the ability to catalytically effect complex reaction cascades in an enantioselective fashion, which continues to expand the boundaries of modern synthetic organic chemistry.

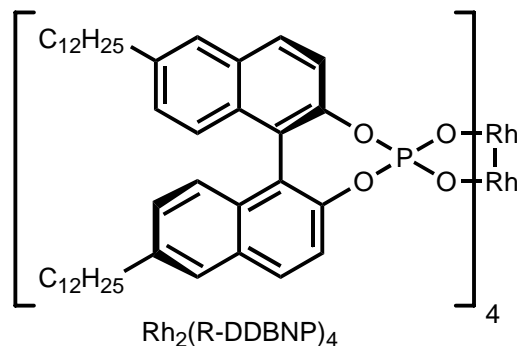
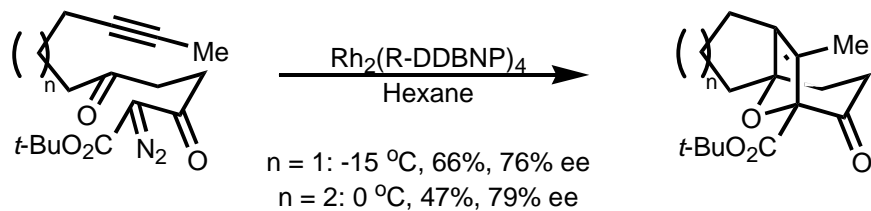
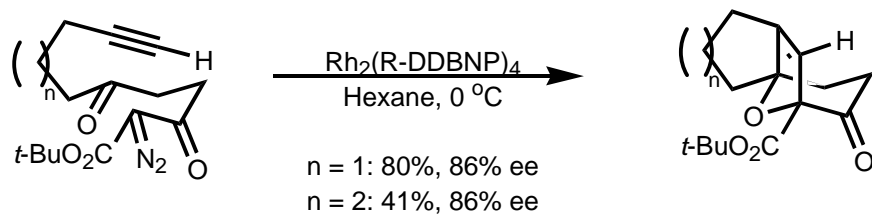
Acknowledgments: The Crimmins Group

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- Anne Marie

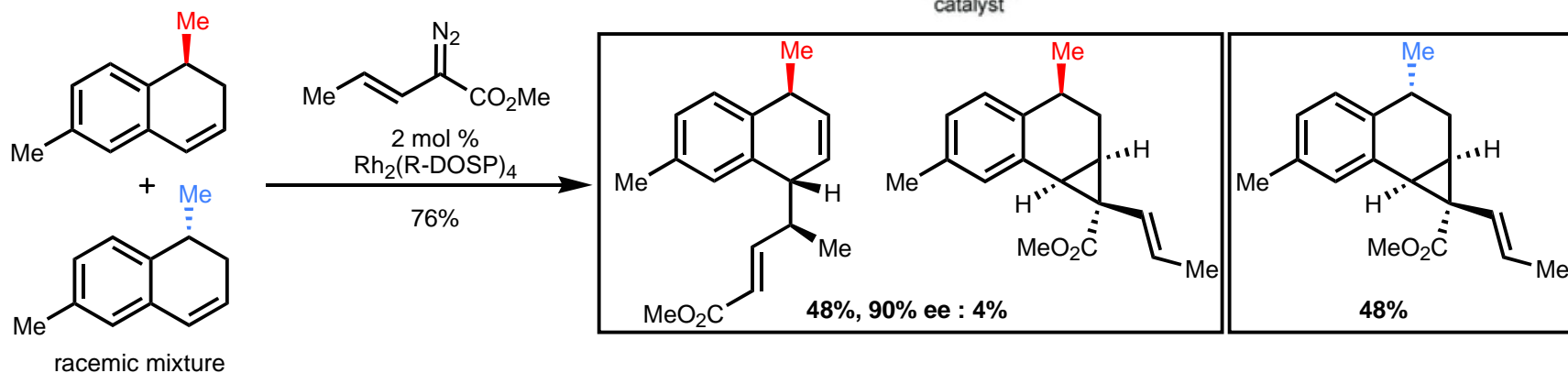
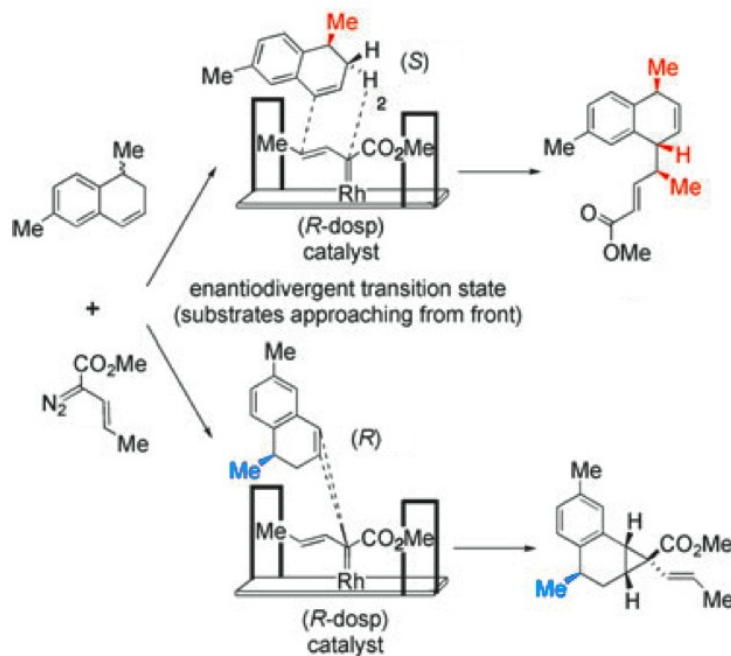
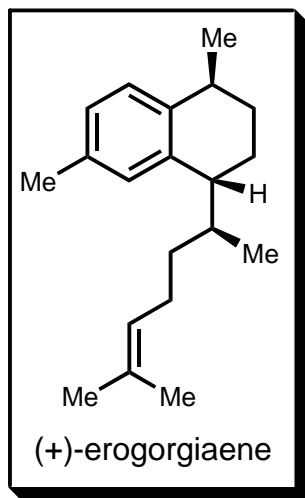


Substituted Alkyne Tethers

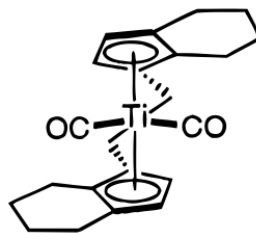


Hodgson, D. M.; Labande, A. H.; Pierard, F.; Expósito Castro, M. A. *J. Org. Chem.* **2003**, *68*, 6153.

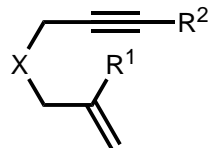
Kinetic Enantiodivergence



Non-Rhodium APKR's



(S,S)-(EBTHI)Ti(CO)₂



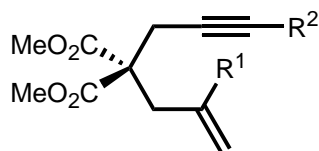
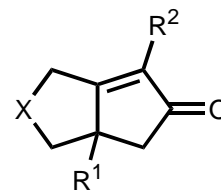
R¹ = H or Me
R² = Me, Ph, *i*-Pr

X = O, NBoc, C(CO₂Me)₂, C(CO₂*t*-Bu)₂

(S,S)-(EBTHI)Ti(CO)₂ 5 - 20 mol %

14 psig CO, PhMe, 12 h, 90 °C

72 - 96%, 70 - 94% ee

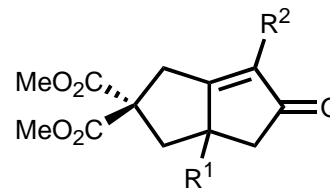


R¹ = H or Me
R² = H or Me

(S)-BINAP 5 - 20 mol %
0.2 mol % Co₂(CO)₈

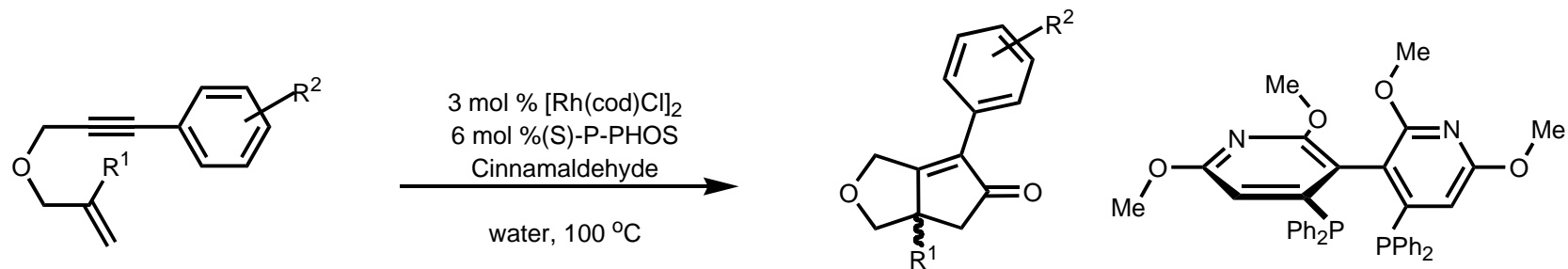
CO, DCE, 19 h, reflux

60%, 90% ee



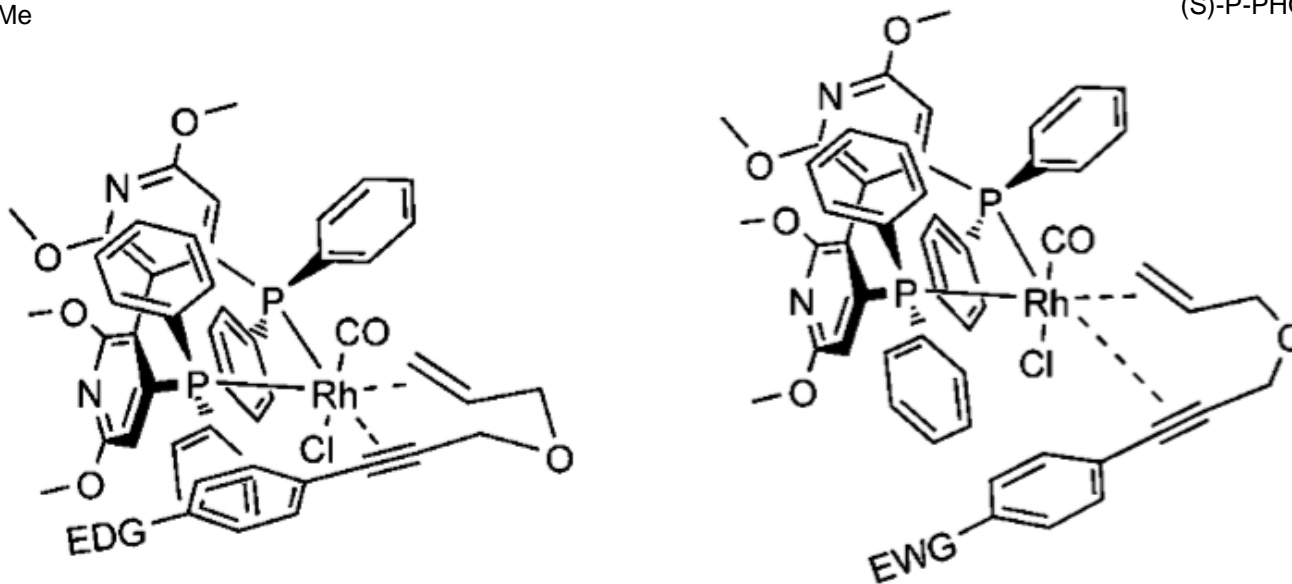
Hicks, F. A.; Buchwald, S. L. *J. Am. Chem. Soc.* **1996**, *118*, 11688. Hicks, F. A.; Buchwald, S. L. *J. Am. Chem. Soc.* **1999**, *121*, 7026. Hiroi, K.; Watanabe, T.; Kawagishi, R.; Abe, I. *Tetrahedron Lett.* **2000**, *41*, 891. Derdau, V.; Laschat, S.; Dix, I.; Jones, P. G. *Organometallics* **1999**, *18*, 3859.

APKR - Electronic Effects



R = H, Me

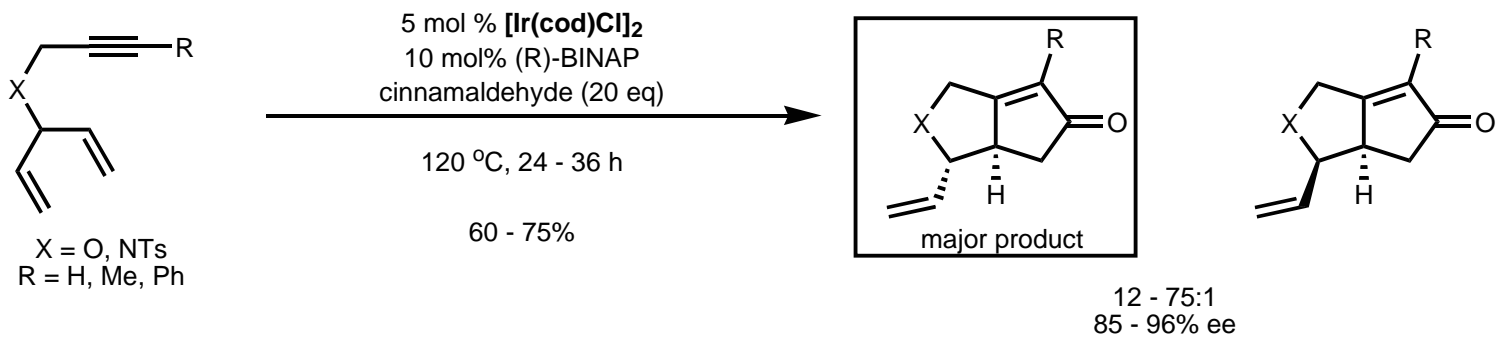
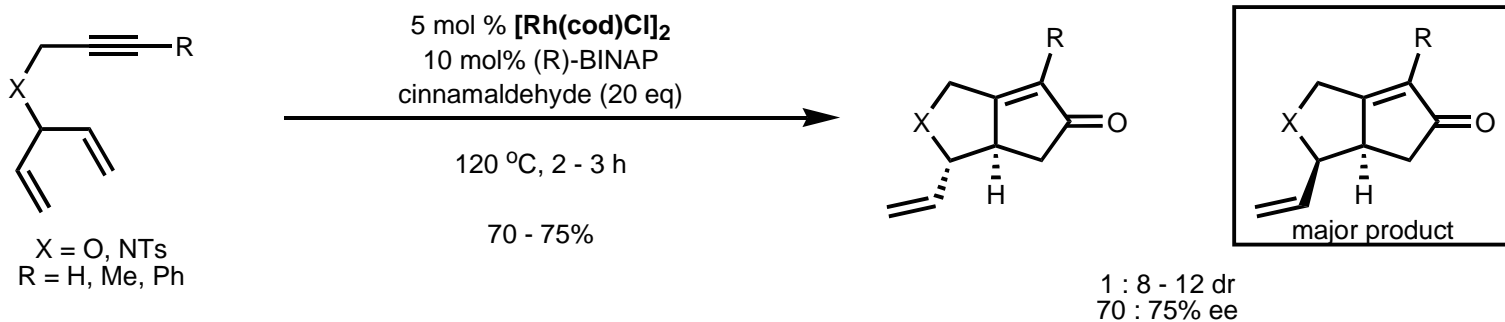
(S)-P-PHOS



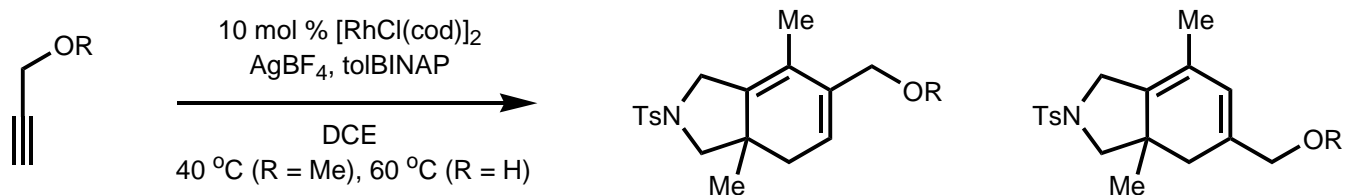
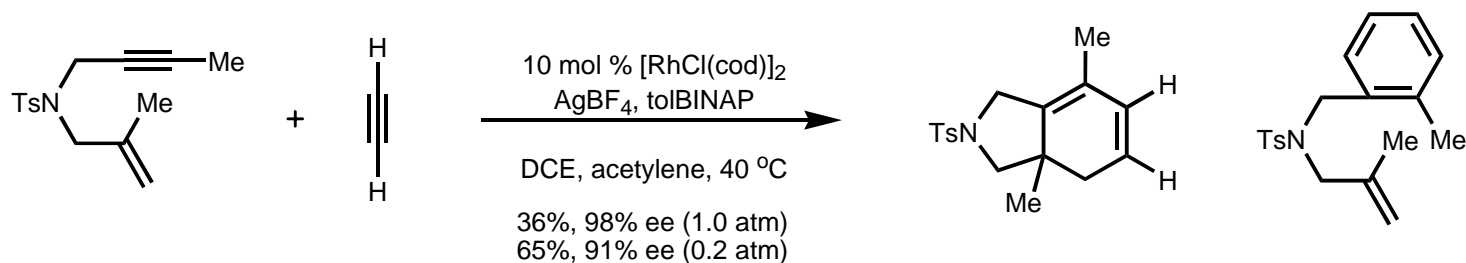
Electron donating groups give higher ee's by drawing the metal and its chiral ligands closer to the electron rich alkyne, allowing for better stereocommunication

Kwong, F. Y.; Li, Y. M.; Lam, H. L.; Qiu, L.; Lee, H. W.; Yeung, C. H.; Chan, K. S.; Chan, A. *Chem. Eur. J.* **2005**, *11*, 3872.

Catalyst Dependent APKR's

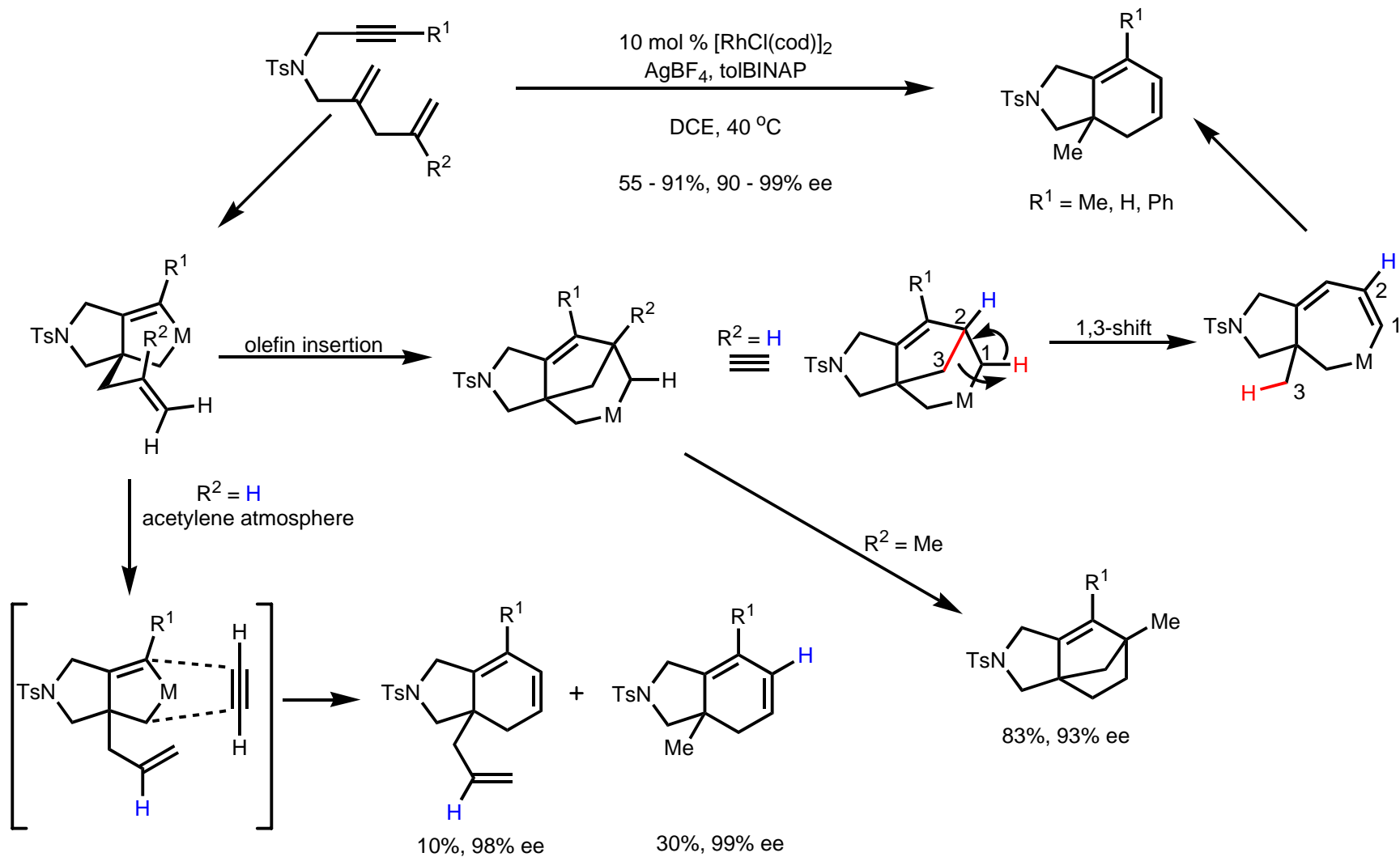


[2 + 2 + 2]’s with Acetylene and Unsymmetrical Alkynes



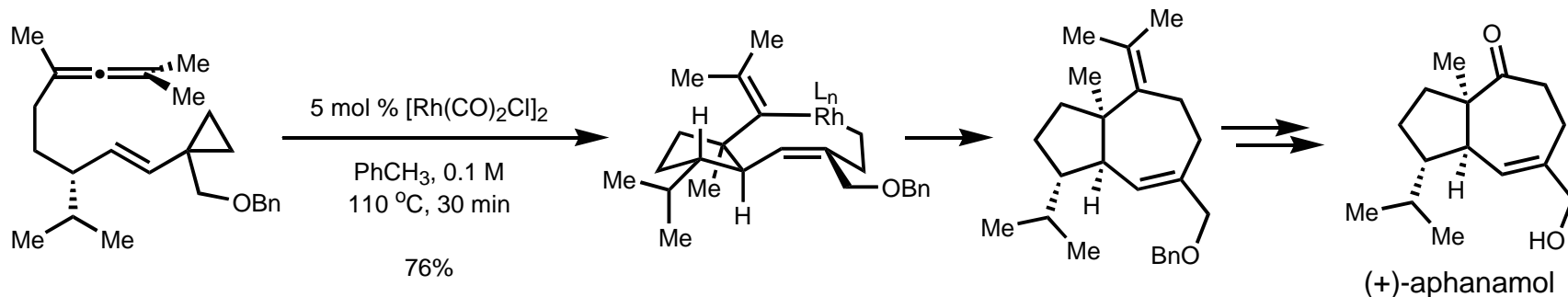
R = Me: 83% 4/1 (92% / 95% ee)
R = H: 63% 7/1 (98% / 97% ee)

1,3-Hydride Shift

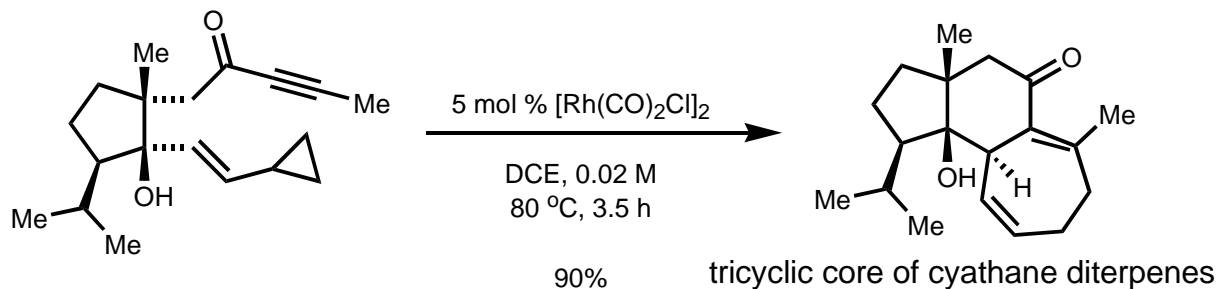


Shibata, T.; Tahara, Y. *J. Am. Chem. Soc.* **2006**, *128*, 11766.

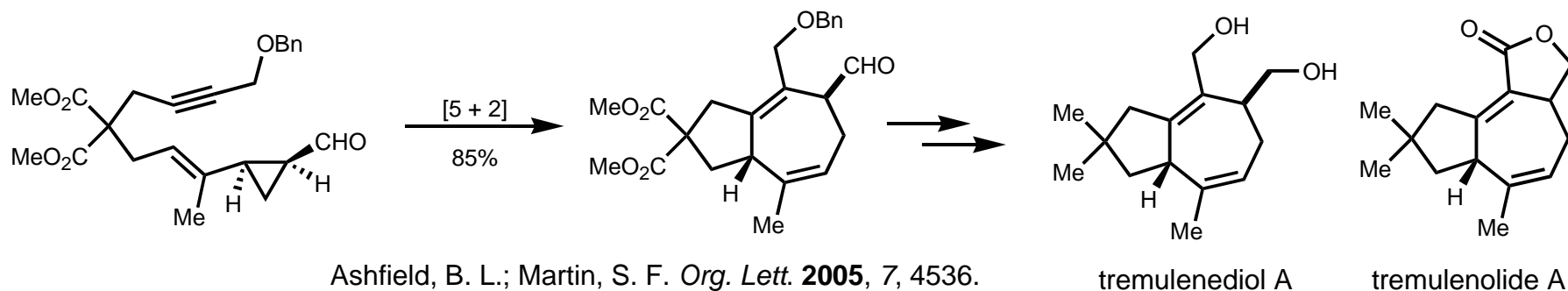
[5 + 2] Cycloaddition in Natural Products Synthesis



Wender, P. A.; Zhang, L. *Org. Lett.*, **2000**, 2, 2323.



Wender, P. A.; Bi, C.; Brodney, M. A.; Gosselin, F. *Org. Lett.* **2001**, 3, 2105.



Ashfield, B. L.; Martin, S. F. *Org. Lett.* **2005**, 7, 4536.